

28-26

Señor Jefe

DIVISION DE NUEVAS CREACIONES

SUPERINTENDENCIA DE INDUSTRIA Y COMERCIO

Santa Fe de Bogotá, D C

SUPERINDUSTRIA Y COMERCIO
Radicacion : 00058244 00002001
Fecha (RFO): 2000-09-07 10:18:02
Tramite : 002 PATENTES D 1 REGISTRO/ 329 COMPLETEN
Dependencia: 2020 DIVISION DE NUEVAS CREACIONES
Folios: 30

Re.: Expediente No. 000-58.244

Solicitud Patentel

SMITHKLINE BEECHAM CORPORATION

Debidamente traducida, adjunto acompaño la copia oficial de la solicitud de patente básica Estadounidense No 60/147-578 la cual le solicito agregar a este expediente para los efectos legales respectivos

De Usted atentamente,

GERMAN CASTILLO GRAU

T.P.#2.978 del C.S.J.

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A TODOS A QUIENES LLEGUE EL PRESENTE:

DEPARTAMENTO DE COMERCIO DE LOS ESTADOS UNIDOS

Oficina de Marcas y Patentes

8 de agosto de 2000

LA PRESENTE CERTIFICA QUE LA ADJUNTA ES UNA COPIA TOMADA DE LOS REGISTROS DE LA OFICINA DE MARCAS Y PATENTES DE LOS ESTADOS UNIDOS, DE LOS DOCUMENTOS DE LA SOLICITUD DE PATENTE ABAJO IDENTIFICADA Y LA CUAL CUMPLE CON LOS REQUISITOS PARA QUE LE SEA CONFERIDA UNA FECHA DE RADICACIÓN SEGÚN 35 USC 111

NÚMERO DE SOLICITUD 60/147,578

FECHA DE RADICACIÓN: 6 de agosto de 1999

Por Autoridad del

COMISIONADO DE MARCAS Y PATENTES

(FIRMA) T WALLACE

Oficial de Certificaciones

(SELLO DORADO DE OBLEA)

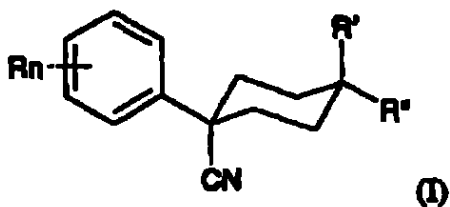
*** * * * ***

METODO PARA PREPARAR ÁCIDOS CICLOHEXAN CARBOXÍLICOS

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3728

REIVINDICACIONES:

- 1 Un proceso para preparar un compuesto de la fórmula (I)



en donde,

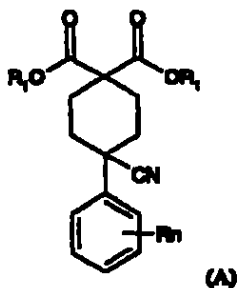
R es halo, alquil C_{1-6} , alquil C_{1-6} sustituido por 1 a 4 halógenos, alcoxi C_{1-6} , alquenil C_{1-6} , $-O-(CH_2)_m$ cicloalquil de 3 a 6 átomos,

n es 1 a 5,

m es 0 a 6, y

R' y R'' son independiente hidrógeno o $CO(O)X$ donde X es hidrogeno o alquil C_{1-6} ,

cuyo proceso consiste en decarboxilar el diácido o diéster de la Fórmula (A)



29 32

en donde R_1 es hidrógeno o un grupo formador de alquil-éster C_{1-6} de 1 a 6 átomos y R y n son lo mismo que para la Fórmula (I)

2 El proceso de la reivindicación 1, en donde el diéster o diácido es combinado con aproximadamente 1 equivalente de una base, aproximadamente 3 equivalentes de agua y aproximadamente 3 equivalentes de una sal álcali en un solvente adecuado calentado a entre aproximadamente 100 a 150°C durante aproximadamente 4 a 8 horas

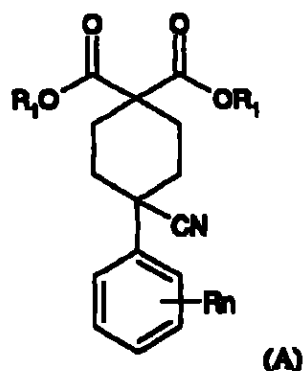
3 El proceso de la reivindicación 1, en donde R_1 es hidrógeno, metil o etil y la base es piridina y la sal es cloruro de litio

4 El proceso según cualquiera de las reivindicaciones 1 a 3, en donde n es R_n y un grupo es sustituido en la posición 3 y el otro grupo es sustituido en la posición 4

5 El proceso de cualquiera de las reivindicaciones 1 a 4, en donde R_1 es metil, uno de R_n es metoxi, $-O-CF_3$, $-O-CHF_2$, o $-O-CH_2CHF_2$ y el otro es cicloalcoxi C_{4-6}

6 El proceso según cualquiera de las reivindicaciones 1 a 5, en donde n en R_n es 2 y uno de 3-ciclopentiloxi y un segundo grupo R_n es 4-metoxi

7 Un compuesto de la fórmula (A)



en donde

R es halo, alquil C_{1-6} , alquil C_{1-6} sustituido con 1 a 4 halogenos, alcoxi C_{1-6} , alqueni C_{1-6} , $O-(CH_2)_m$ cicloalquil de 3 a 6 átomos,

n es 1 a 5,

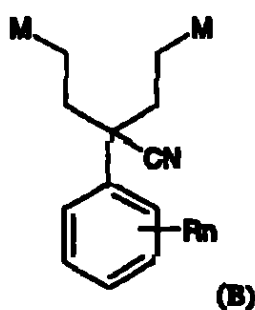
m es 0 a 6,

R₁ es hidrógeno o un grupo formador de alquil éster C₁₋₆ de 1 a 6 átomos de carbono,

8 Un compuesto según la reivindicación 8 en donde n en R_n es 2 y R_n es metoxi, -O-CF₃, -O-CHF₂, o -O-CH₂CHF₂ y el otro es cicloalquiloxiC₄₋₆

9 Un compuesto según cualquiera de las reivindicaciones 8 o 9, en donde n en R_n es 2 y uno es 3-ciclopentiloxi y un segundo grupo R_n es 4-metoxi

10- Un compuesto de la Formula (B)



en donde

R es halo, alquil C₁₋₆, alquil C₁₋₆ sustituido por 1 a 4 halógenos, alcoxi C₁₋₆, alqueni C₁₋₆, -O-(CH₂)_mcicloalquil de 3 a 6 átomos,

n es 1 a 5,

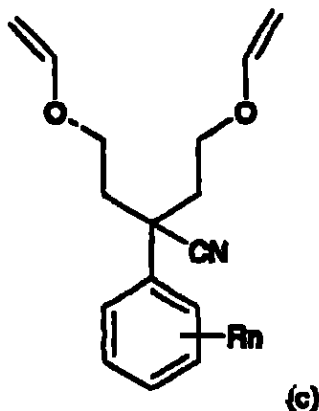
m es 0 a 6, y

M es OH, un grupo hidroxil activado, o halo

12 Un compuesto según la reivindicación 11, en donde n en R_n es 2 y R_n es metoxi, -O-CF₃, -O-CHF₂, o -O-CH₂CHF₂ y el otro es cicloalquiloxiC₄₋₆

13 Un compuesto según la reivindicación 11 o 12, en donde n en R_n es 2 y uno es 3-ciclopentiloxi y el segundo grupo R_n es 4-metoxi

14 Un compuesto de la Fórmula (C)



en donde

R es halo, alquil C_{1-6} , alquil C_{1-6} sustituido por 1 a 4 halógenos, alcoxi C_{1-6} , alqueni C_{1-6} , $-O-(CH_2)_m$ cicloalquil de 3 a 6 átomos,

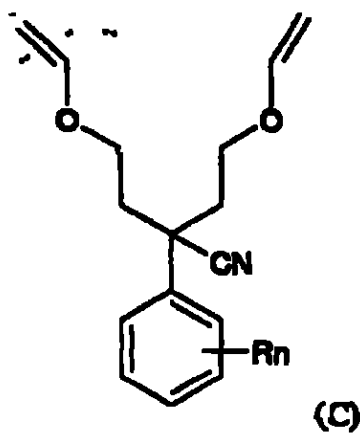
n es 1 a 5,

m es 0 a 6

15 Un compuesto según la reivindicación 14, en donde n en R_n es 2 y R_n es metoxi, $-O-CF_3$, $-O-CHF_2$, o $-O-CH_2CHF_2$ y el otro es cicloalquiloxi C_{4-6}

16 Un compuesto según la reivindicación 14, en donde n en R_n es 2 y uno es 3-ciclopentiloxi y el segundo grupo R_n es 4-metoxi

17 Un proceso para preparar un compuesto de la Fórmula (C)



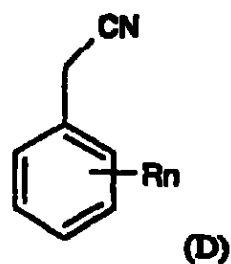
en donde

R es halo, alquil C₁₋₆, alquil C₁₋₆ sustituido por 1 a 4 halógenos, alcoxi C₁₋₆, alquenil C₁₋₆, -O-(CH₂)_mcicloalquil de 3 a 6 átomos,

n es 1 a 5,

m es 0 a 6,

que consiste en tratar el nitrilo de la Formula (D)



con 2-cloroetil vinil éter y una base fuerte,

donde, en la Fórmula (D),

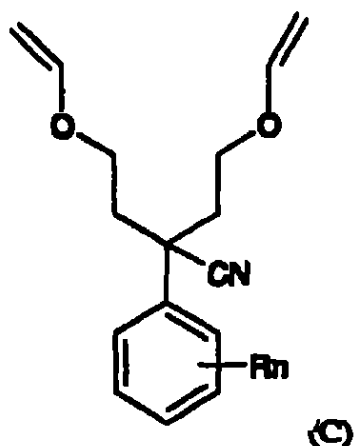
R es halo, en donde alquil C₁₋₆, alquil C₁₋₆ sustituido por 1 a 4 halogenos, alcoxi C₁₋₆, alquenil C₁₋₆, -O-(CH₂)_mcicloalquil de 3 a 6 átomos,

n es 1 a 5,

m es 0 a 6

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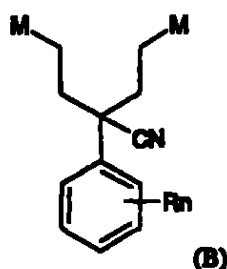


en donde R y n son R es halo, en donde alquil C₁₋₆, alquil C₁₋₆ sustituido por 1 a 4 halogenos, alcoxi C₁₋₆, alquenil C₁₋₆, -O-(CH₂)_mcicloalquil de 3 a 6 átomos,

nes 1 a 5,

mes 0 a 6

en un compuesto de la Fórmula (B)

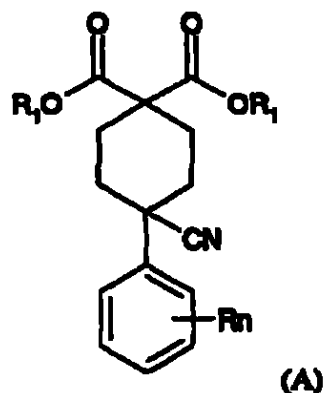


en donde M es OH ,

b convertir el grupo hidroxil de la Fórmula (B) en un compuesto de la Fórmula (B), donde M es un tosilato, mesilato o un triflato

c convertir el tosilato, mesilato o triflato en la Fórmula (B) en un compuesto de la Fórmula (B), donde M es halo⁺

d tratar el compuesto dihalo con malonato de dialquil para obtener un compuesto de la Fórmula (A)



donde R₁ es alquil más bajo,

e opcionalmente, saponificar el diéstero de la Fórmula (A) para obtener un compuesto de la Fórmula (A), donde R₁ es hidrógeno, y

f descarboxilar un compuesto de la Fórmula (A), donde R₁ es hidrógeno o alquil C₁₋₆ para obtener un compuesto para la Fórmula (I), donde uno de R' es hidrógeno y el otro es CO(O)X, donde X es alquil C₁₋₆ o hidrógeno

19 El proceso de la reivindicación 18, en donde n en R_n es 2 y R_n es metoxi, -O-CF₃, -O-CHF₂, o -O-CH₂CHF₂ y el otro es cicloalquiloxiC₄₋₆, M es tosilato, y de allí en adelante yodo, y R₁ es metil o etil

20 Un compuesto según la reivindicación 18 o 19, en donde n en R_n es 2 y uno es 3-ciclopentiloxi y el segundo grupo R_n es 4-metoxi

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THE UNITED STATES OF AMERICA

TO ALL TO WHOM THESE PRESENTS SHALL COME:

UNITED STATES DEPARTMENT OF COMMERCE

United States Patent and Trademark Office

August 08, 2000

THIS IS TO CERTIFY THAT ANNEXED HERETO IS A TRUE COPY FROM THE RECORDS OF THE UNITED STATES PATENT AND TRADEMARK OFFICE OF THOSE PAPERS OF THE BELOW IDENTIFIED PATENT APPLICATION THAT MET THE REQUIREMENTS TO BE GRANTED A FILING DATE UNDER 35 USC 111.

APPLICATION NUMBER: 60/147,578 /

FILING DATE: August 06, 1999 /



By Authority of the
COMMISSIONER OF PATENTS AND TRADEMARKS

T. Wallace
T. WALLACE

Certifying Officer

"EXPRESS MAIL CERTIFICATE"

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I Herby Certify That This Paper Or Fee Is Being Deposited With The United States Postal Service "Express Mail Post Office To Addressee" Service Under 37 CFR 1.10 On The Date Indicated Above And Is Addressed To: Assistant Commissioner for Patents, Washington, D.C. 20231.

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PROVISIONAL APPLICATION COVER SHEET

This is a request for filing a PROVISIONAL APPLICATION for PATENT under 37 CFR 1.53(c).

Docket No.

P50972P

INVENTOR(s) / APPLICANT(s)

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TITLE OF THE INVENTION (280 characters max)

METHOD FOR PREPARING CYCLOHEXANE CARBOXYLIC ACIDS

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ENCLOSED APPLICATION PARTS (check all that apply)

☒ Specification	Number of Pages	17	☐ Small Entity Statement
☐ Drawings	Number of Sheets		☒ Other (specify)
			1 page abstract

METHOD OF PAYMENT OF FILING FEES FOR THIS PROVISIONAL APPLICATION FOR PATENT

☒ The Commissioner is hereby authorized to charge filing fees and credit Deposit Account No. 19-2570

PROVISIONAL FILING
FEE AMOUNT (\$)

\$150.00

The invention was made by an agency of the United States Government or under a contract with an agency of the United States Government.

☒ No.

☐ Yes, the name of the U.S. Government agency and the Government contract number are: _____

Respectfully submitted,

Signature:

James M. Kanagy
James M. Kanagy

Date.

Registration No.

06 Aug 1999
29,554

o Additional inventors are being named on separately numbered sheets attached hereto.

PROVISIONAL APPLICATION FILING ONLY

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37-22
6/0Method For Preparing Cyclohexane Carboxylic AcidsArea of the Invention

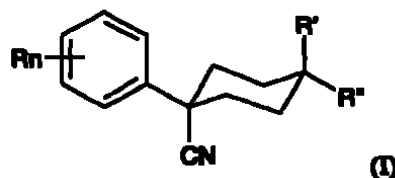
This invention relates to a method for preparing 4-substituted-4-cyanocyclohexanecarboxylic acids. Exemplary compounds are useful as PDE 4 inhibitors.

5 Background of the Invention

The process of this invention relates to making compounds which are useful in treating diseases modulated by the isoforms of the phosphodiesterase 4 enzyme. The novel intermediates and processes of this invention are useful in making acids which are known PDE 4 inhibitors. They are useful for, among other things, treating pulmonary diseases such as chronic obstructive pulmonary disease (COPD) and asthma. The compounds which are prepared by the methods of this invention are described in, for example U.S. patent 5,554,238 issued 03 September, 1996. That patent is incorporated here by reference in full. Those compounds, particularly the 4-cyanocyclohexanecarboxylic acids, have marked effects on neutrophil activity, inhibiting neutrophil chemotaxis and degranulation *in vitro*. In animal models, those compounds reduce neutrophil extravasation from the circulation, pulmonary sequestration and the edematous responses to a number inflammatory insults *in vivo*. They have been found to be useful in treating COPD in humans, and possibly in other mammalian species which suffer from COPD.

Summary of the Invention

20 In a first aspect, this invention relates to a process for preparing a compound of formula (I)



where

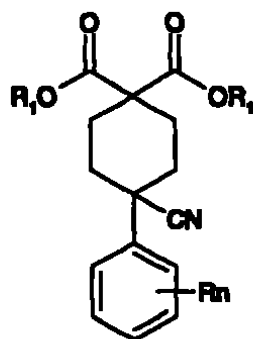
25 R is halo, C₁-alkyl, C₁-alkyl substituted with 1 to 4 halogens, C₁-alkoxy, C₁-alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons,

n is 1-5,

m is 0 - 6, and

30 R' and R'' are independently hydrogen or CO(O)X where X is hydrogen or C₁-alkyl,

which process comprises decarboxylating the diester or diacid of Formula (A)

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(4)

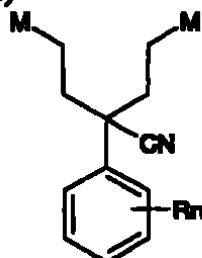
(A)

where R_1 is hydrogen or a C_{1-6} alkyl-ester forming group and R and n are the same as for Formula (I)

In a further aspect this invention relates to a compound of formula (A) per se.

- 5 In a third aspect this invention relates to preparing certain other intermediates that are useful in preparing the diester or di-acid of Formula (A), and the intermediates themselves, i.e.

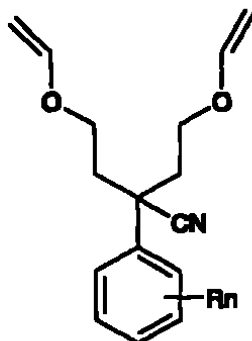
a compound of Formula (B)



(B)

- 10 wherein R and n are the same as in Formula (I) and M is OH, an activated hydroxyl group, or halo; and

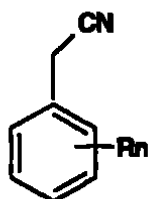
a compound of Formula (C)



(C)

wherein R and n are the same as in Formula (I).

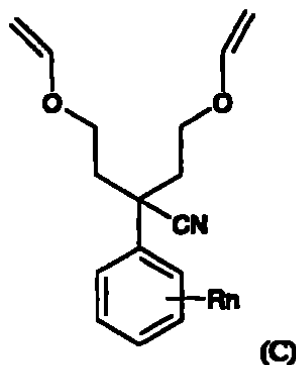
- 15 In yet another aspect, the invention provides a method for making a compound of Formula (C) by treating the nitrile of formula (D)

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(D)

where R and n are the same as defined above, with 2-chloroethyl vinyl ether and a strong base.

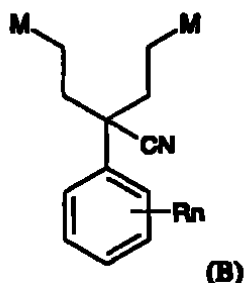
- This invention also provides a method for preparing a compound of Formula (I)
 5 which comprises
 a. converting the vinylethyl ether of Formula (C)



(C)

- 10 wherein R and n are halo, C₁-alkyl, C₁-alkyl substituted with 1 to 4 halogens, C₁-alkoxy, C₁-alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons, n is 1-5;
 m is 0 - 6,
 to a compound of Formula (B)

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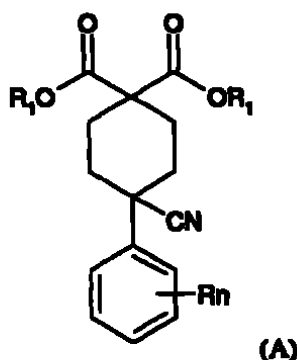


(B)

where M is OH,

- b. converting the hydroxyl group of Formula (B) to a compound of Formula (B)
 20 where M is a tosylate, mesylate or a triflate,
 c. converting the tosylate, mesylate or triflate in Formula (B) to a compound of Formula (B) where M is halo,

d treating the di-halo compound with dialkyl malonate to obtain a compound of Formula (A)



5 where R_1 is lower alkyl,

e. optionally saponifying the diester of Formula (A) to obtain a compound of Formula (A) where R_1 is hydrogen, and

f decarboxylating a compound of Formula (A) where R_1 is hydrogen or C_{1-6} alkyl to obtain a compound for Formula (I) where one of R' is hydrogen and the other is $CO(O)X$ where X is C_{1-6} alkyl or hydrogen

Specific Embodiments of the Invention

This invention provides a method for preparing cyclohexanoic acids. In particular it provides an alternative means for preparing the cyclohexanoic acids disclosed in U.S. patent 5,554,238 where the 4-position on the cyclohexane ring has a CN group

15 "Halo" as used herein includes fluoro, chloro, bromo, and iodo. "Halide" includes fluoride, chloride, bromide and iodide.

For all of the compounds disclosed herein, a preferred embodiment is one where there are two R groups, i.e. n is 2. Most preferred are those compounds where one R group is at the 3 position and the second R group is on the 4 position of the benzene ring. More particularly it is preferred that each R group be independently C_{4-6} cycloalkyloxy or C_{1-2} alkoxy unsubstituted or substituted by 1 or more halogens. More preferred are methoxy, C_{1-2} alkoxy substituted by up to 3 fluoro atoms, cyclopropylmethoxy or cyclopentylloxy. The more preferred R groups are those wherein the 4-position R group is methoxy, $-O-CF_3$, $-O-CHF_2$, or $-O-CH_2CHF_2$, and the 3-position R group is cyclopropylmethoxy or cyclopentylloxy

In Formula (A) the most preferred R_1 groups are hydrogen, methyl or ethyl

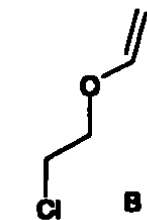
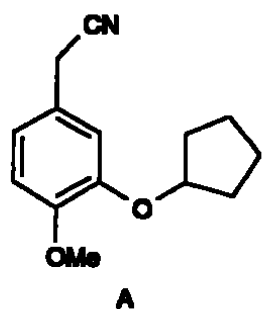
In Formula (B) the most preferred M groups are OH, tosyl and iodo

The most preferred product of the process of this invention are those compounds which have a 3-cyclopentylloxy-4-methoxyphenyl substitution pattern.

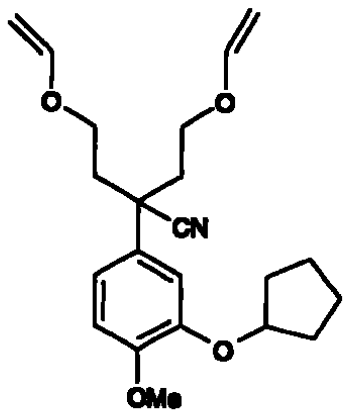
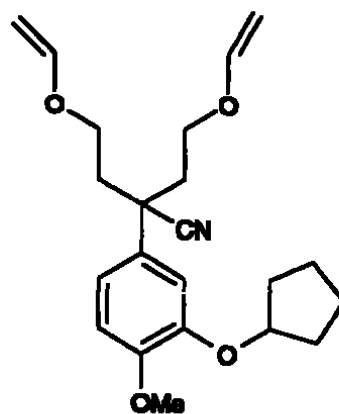
30 Reaction Scheme I provides a diagrammatic overview of the intermediates and chemistries employed in this invention.

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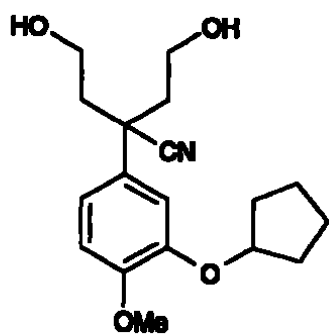
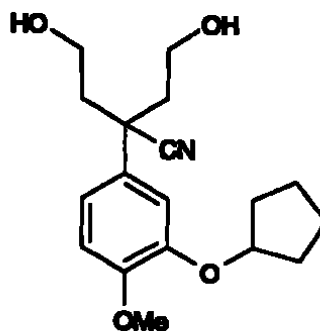
Scheme I



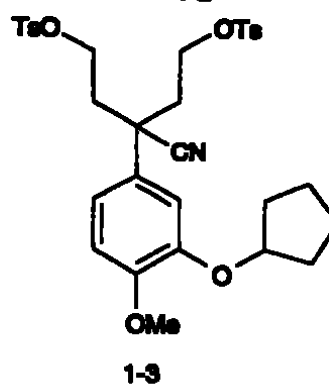
Toluene
NaNH₂

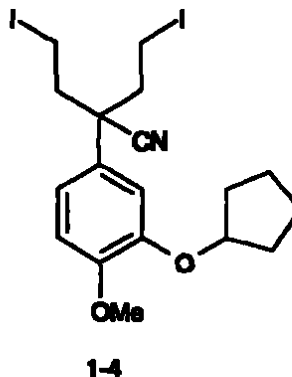
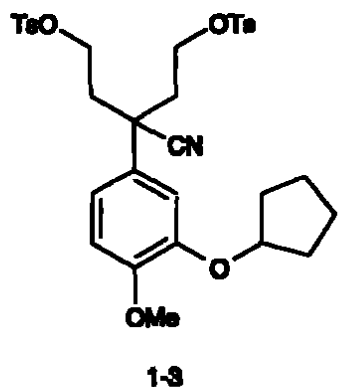


aqueous HCl
80-85 deg. C

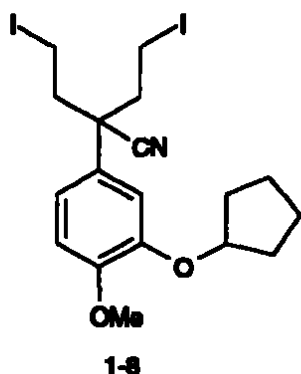


TsCl
pyridine

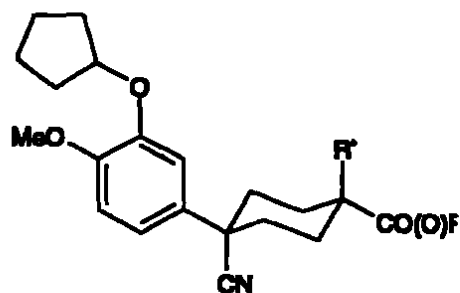
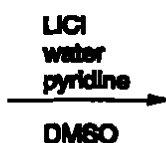
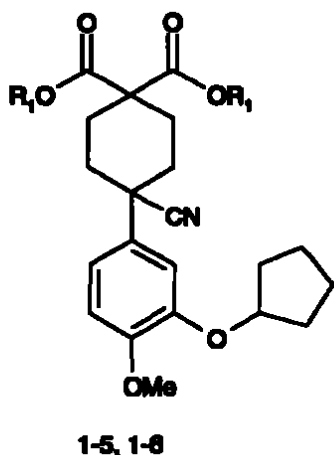
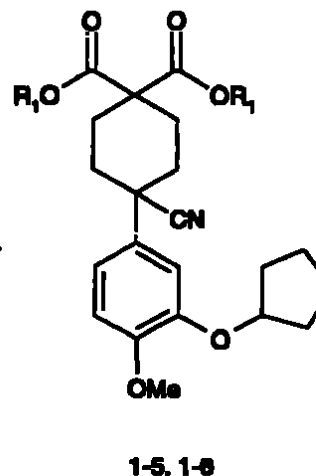
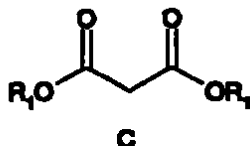




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- 5 The starting material 3-cyclopentyloxy-4-methoxybenzeneacetonitrile is a known compound. See for example U.S. patent 5,449,686 The 2-chloroethylvinyl ether is commercially available (Aldrich) To effect the reaction, a strong base is charged to a reaction vessel containing a suitable non-polar solvent to which the vinyl ether is added. This mixture is heated to between about 30 to 70 °C and charged with the
- 10 benzeneacetonitrile (A) pre-dissolved in the same solvent as the base and the vinyl ether Toluene is a preferred solvent. A preferred base is sodium amide The amount of base is

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equivalent, on a molar basis, to that of the vinyl ether. Both are used in about a three-fold excess relative to the substrate. After the benzenecetonitrile has been charged to the reaction flask, the solution is further heated to around 80 °C more or less. Usually the reaction is complete in about 30 minutes to 2 hours. The product (1-1) is isolated using standard procedures.

The bis 2-hydroxyethyl compound (1-2) is prepared by treating the vinyl ether moiety prepared as per the preceding paragraph with a strong mineral acid in an aqueous solvent. For example water can be added to the 2-(ethenyloxy)ethyl compound (1-1), heating that combination to about 70-90 °C and then adding a molar excess of a mineral acid such as HCl or the like. A preferred set of conditions is one where the 2-(ethenyloxy)ethyl is treated with water and heated to about 80 °C more or less followed by the addition of a 50% molar excess of concentrated HCl. Under these conditions the reaction is complete in 5 - 20 minutes.

To obtain the halogenated compound 1-4, the diol is converted to a group which can be displaced by a halide ion. For example the diol can be converted to a tosylate, mesylate, or the like, by treating the diol with reagents and under conditions which form the tosylate, etc. By way of example the diol is dissolved in an organic solvent and treated with an excess of *p*-toluenesulfonyl chloride at room temperature for 3- 7 hours. Preferably the reaction is run in pyridine with about a 2.5 molar excess of the *p*-toluenesulfonyl chloride.

This tosylate (or mesylate, triflate, etc) (1-3) is converted to the di-halo 1-4 by dissolving it in a polar aprotic solvent, and adding a weak base and a halide salt. This mixture is heated to reflux for a number of hours, for example overnight. A preferred solvent is acetone or dimethyl formamide. A preferred halide salt is sodium or lithium iodide though other sodium or potassium salts of fluormine, chlorine and bromine can be used as well. A 2 to 6-fold excess of the halide salt is preferred. Refluxing overnight usually effects completion of the reaction.

Forming the cyclohexane dicarboxylates or diacids 1-5 and 1-6 is effected by charging the di-halo compound (1-4) to a solution of a dialkyl malonate or malonic acid and a weak base in a dipolar aprotic solvent. This slurry is stirred for an extended period of time at an elevated temperature, for example overnight. More specifically sodium or potassium carbonate is combined with the likes of dimethyl malonate in a solvent such as dimethylformamide. Then the di-halo 1-4 is added and the resulting slurry is stirred overnight at about 75-95 °C or so. The malonate is added in about a 1:1 molar ratio to that of the di-halo compound.

The diester may be saponified to give the diacid, though this step is not illustrated in Scheme 1. This is accomplished by treating the diester with an aqueous base in a water-immiscible solvent. For example the diester is charged to a reaction vessel containing the likes

of tetrahydrofuran to which is added water and an alkali hydroxide base such as lithium hydroxide. This solution is heated at reflux for a number of hours, for example overnight.

Decarboxylating the diester or diacid to get the mono-ester or mono-acid is accomplished by dissolving the diester in the likes of dimethylsulfoxide, adding about an equivalent of a base such as pyridine, about 3 equivalents of water and about 3 equivalents of a salt such as lithium chloride. This solution is stirred for several hours at 100 to 150 °C or thereabouts for 4-8 hours. Product is extracted from an acidified aqueous solution and further purified by conventional means. The product is a mixture of *cis* and *trans* isomers in about a 1:1 ratio. The *cis* form of the ester or acid can be enriched by dissolving a mixture of isomers in a lower alkanol and treating that solution with the alkali metal salt of the alkanol. A preferred alkanol is *t*-butanol and a preferred alkali metal salt is potassium *t*-butanol. The acid may be obtained by saponifying the ester using a base and then acidifying the resulting salt with using a mineral acid, for example

The following examples are provided to illustrate the invention. These illustrative examples are not intended to limit the claimed invention in any fashion

Examples

Example 1

Preparation of 3-(Cyclopentyloxy)- α,α -bis[2-(ethoxyloxy)ethyl]-4-methoxybenzeneacetonitrile

A 1 L flask was charged with 150 mL of toluene, sodium amide (16.5 g, 0.38 mole, 2.9 equivalents), and 2-chloroethyl vinyl ether (41.9 g, 0.39 mole, 3.0 equivalents). The suspension was heated to 50 °C, then charged with a solution of 3-cyclopentyloxy-4-methoxybenzeneacetonitrile (30 g, 0.13 mole, 1.0 equivalents) in 150 mL of toluene. The reaction was then carefully heated to 80 °C. The progress was followed by HPLC (acetonitrile/0.1 N aqueous ammonium acetate at 65/35, 15 cm Beckman ODS Ultrasphere, 2 mL/min, 215 nm UV). After 60 minutes, the solution was poured into 1 L of water and 300 mL of *t*-butyl methyl ether. The layers were separated, the organic layer washed with water, then brine. The solvent was removed under reduced pressure to give a brown oil (52.1 g). The captioned compound was isolated by column chromatography (230-400 mesh silica gel, 10/1 hexane/ethyl acetate).

Mass spectrometry gave $m/z = 372 (M+H^+)^+$

¹H NMR, 300 MHz, CDCl₃, δ ppm) δ 1.55-1.65 (m, 2H, ring CH₂), δ 1.70-2.00 (m, 6H, ring CH₂'s), δ 2.2-3.5 (m, 4 H, (CH₂)₂CCN), δ 3.50-3.85 (m, 4H, CH₂O), δ 3.85 (s, 3H, OCH₃), δ 3.95-4.10 (m, 4H, CH₂ alkene), δ 4.8 (m, 1H, ring CH), δ 6.30-6.40 (m, CH, alkene), δ 6.85-7.0 (m, 3H, aromatic)

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Example 2**Preparation of 3-(Cyclopentylloxy)- α,α -bis(2-hydroxyethyl)-4-methoxybenzeneacetonitrile****Purified 3-(cyclopentylloxy)- α,α -bis[2-(ethenylloxy)ethyl]-4-**

methoxybenzeneacetonitrile (5 g, 13.5 mmol) was treated with water (50 mL) and heated to 80 °C with rapid stirring. Concentrated hydrochloric acid (1.85 mL, 22.2 mmol) was added and stirring was continued for 10 minutes. The solution was poured into ice water (50 mL) and methylene chloride (50 mL). The layers were separated, and the aqueous layer was extracted once with methylene chloride. The combined organic layers were washed with water, then brine, and concentrated to a light yellow oil in quantitative yield. Structure and purity were confirmed by ¹H NMR.

(¹H NMR, 300 MHz, CDCl₃, δ ppm) δ 1.55-1.70 (m, 2H, ring CH₂), δ 1.75-1.95 (m, 6H, ring CH₂'s), δ 2.10-2.40 (m, 4H, (CH₂)₂CCN), δ 3.55-3.85 (m, 4H, CH₂O), 3.88 (s, 3H, OMe), δ 4.8 (m, 1H, ring CH), δ 6.80-7.00 (m, 3H, aromatic)

Example 3

Preparation of 3-(Cyclopentylloxy)-4-methoxy- α,α -bis[2-[(4-methylphenyl)sulfonyl]oxy]ethyl]benzeneacetonitrile

A 250 mL flask was charged with 3-(cyclopentylloxy)- α,α -bis(2-hydroxyethyl)-4-methoxybenzeneacetonitrile (6.4 g, 20.1 mmol), pyridine (65 mL), and *p*-toluenesulfonyl chloride (9.56 g, 50.2 mmol). The solution warmed slightly (exothermic), then was stirred at room temperature for 5 hours. The reaction was deemed complete by HPLC (acetonitrile/0.1 N aqueous ammonium acetate at 65/35, 15 cm Beckman ODS Ultrasphere, 2 mL/min, 215 nm UV). The reaction was poured into 100 mL of 5% HCl and 50 mL of methylene chloride. The layers were separated, and the organic layer was washed with 5% HCl until neutral. The neutralized organic layer was then washed once with brine and concentrated. The captioned compound was isolated as a white solid by crystallization from ethanol and *t*-butyl methyl ether.

Mass spectrometry gave m/z = 645 ($M+NH_4^+$)⁺

(¹H NMR, 300 MHz, CDCl₃, δ ppm) δ 1.55-1.70 (m, 2H, ring CH₂), δ 1.72-2.0 (m, 6H, ring CH₂'s), δ 2.20-2.45 (m, 4H, (CH₂)₂CCN), 2.45 (s, 3H, ar-CH₃), 3.85 (s, 3H, OMe), δ 3.85-4.28 (m, 4H, CH₂O), δ 4.75 (m, 1H, ring CH), δ 6.75-7.75 (11H indicated, aromatic)

Example 4**Preparation of 3-(Cyclopentylloxy)- α,α -bis(2-iodoethyl)-4-methoxybenzeneacetonitrile**

A 250 mL flask was charged with 3-(cyclopentylloxy)-4-methoxy- α,α -bis[2-[(4-methylphenyl)sulfonyl]oxy]ethyl]benzeneacetonitrile (5.0 g, 7.97 mmol), acetone (75 mL), and sodium bicarbonate (50 mg). This solution was stirred well while sodium iodide (5.98 g, 39.9 mmol) was added, then heated to reflux overnight. The reaction was poured into aqueous ammonium chloride and sodium bisulfite, then extracted with *t*-butyl methyl ether.

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The organic layer was dried over sodium sulfate, then concentrated to a clear colorless oil. The captioned compound was crystallized from *t*-butyl methyl ether and hexanes to give a white solid.-

Mass spectrometry gave $m/z = 540 (M+H^+)^+$

- 5 CHN analysis calculated for $C_{18}H_{23}NO_2I_2$ (539.12) C 40.10, H 4.30, N 2.60; found C 40.06, H 4.30, N, 2.45

(1H NMR, 300 MHz, $CDCl_3$, δ ppm) δ 1.55-1.70 (m, 2H, ring CH_2), δ 1.75-2.05 (m, 6H, ring CH_2 's), δ 2.35-2.62 (m, 4H, $(CH_2)_2CCN$), δ 2.8 (m, 2H, CH_2I), δ 3.15 (m, 2H, CH_2I), 3.87 (s, 3H, OMe), δ 4.8 (m, 1H, ring CH), δ 6.85 (s, 1H, aromatic), δ 7.18-7.30 (m, 2H, aromatic)

Example 5

Preparation of Dimethyl 4-cyano-4-[3-(cyclopentyloxy)-4-methoxyphenyl]-1,1-cyclohexanedicarboxylate and Diethyl 4-cyano-4-[3-(cyclopentyloxy)-4-methoxyphenyl]-1,1-cyclohexanedicarboxylate

- 15 A 125 mL flask was charged with 60 mL of dimethylformamide, potassium carbonate (5.02 g, 36.4 mmol, 3.6 equivalents), dimethylmalonate (1.33 g, 10.1 mmol, 1.0 equivalents), and then 3-(cyclopentyloxy)- α,α -bis(2-iodoethyl)-4-methoxybenzonnitrile (6.0 g, 11.1 mmol, 1.1 equivalents). The slurry was stirred at 80 °C overnight. The completed reaction was poured into 50 mL of water and 50 mL of *t*-butyl methyl ether. The organic layer was
- 20 extracted three times with water, then once with brine. The product was isolated by column chromatography (Flash silica [230-400 mesh], 80/20 hexanes/ethyl acetate). Alternatively it was crystallized from hexanes/ethyl acetate (3/1) to give white crystals. The diethyl derivative was prepared using the same procedure.

Dimethyl ester:

- 25 Mass spectrometry gave $m/z = 416 (M+H^+)^+$
- CHN analysis calculated for $C_{23}H_{29}NO_6$ (539.12) C 66.49, H 7.04, N 3.37, found C 66.24, H 6.94, N, 3.33
- (1H NMR, 300 MHz, $CDCl_3$, δ ppm) δ 1.55-1.67 (m, 2H, ring CH_2), δ 1.75-2.60 (m, 14H, ring CH_2 's), 3.75 (s, 3H, CO_2Me), 3.78 (s, 3H, CO_2Me), 3.82 (s, 3H, OMe), δ 4.8 (m, 1H, ring CH), δ 6.80-7.02 (m, 3H, aromatic)

Diethyl ester

- Mass spectrometry gave $m/z = 444 (M+H^+)^+$
- mp 74.0-74.5
- (1H NMR, 300 MHz, $CDCl_3$, δ ppm) δ 1.20-1.38 (m, 6H, ethyl CH_3), δ 1.50-2.60 (m, 16H, ring CH_2 's), 3.85 (s, 3H, OMe), δ 4.65-4.85 (m, 4H, ethyl CH_2), δ 4.8 (m, 1H, ring CH), δ 6.80-7.05 (m, 3H, aromatic).

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Example 6**Hydrolysis of the diethyl ester to the diacid**

A 25 mL flask was charged with tetrahydrofuran (5 mL), the diethyl ester (SB 220523, 0.5 g, 1.13 mmol, 1.0 equivalent), water (5 mL), and lithium hydroxide monohydrate (0.95 g, 22.6 mmol, 20 equivalents). The solution was stirred at reflux for 18 hours. The reaction was deemed complete by HPLC (15 cm Supelco LC-ABZ, 40/60/0.1 [acetonitrile/water/TFA], 1.5 mL/min., 215 nm UV). The reaction solution was then cooled and diluted with 10% HCl and *t*-butyl methyl ether. The layers were separated and the aqueous layer was washed once with *t*-butyl methyl ether. The organic layers were combined and washed with water and then brine. The solution was then concentrated to a tan solid. Water was removed by reconcentrating once with acetonitrile. The crude product was obtained in about 90% yield. The crude product showed residual ethyl ester (< 5%) by HPLC and ¹H NMR.

(¹H NMR, 300 MHz, CDCl₃, δ ppm) δ 1.55-1.70 (m, 2H, ring CH₂), δ 1.75-2.35 (m, 12H, ring CH₂'s), δ 2.52-2.63 (m, 2H, ring CH₂), 3.85 (s, 3H, OMe), δ 4.8 (m, 1H, ring CH), δ 6.8-7.0 (m, 3H, aromatic)

Example 7**Decarboxylation of the diester: Ethyl 4-cyano-4-(3-cyclopentyloxy-4-methoxyphenyl)-cyclohexanecarboxylate**

A 100 mL flask was charged with dimethylsulfoxide (35 mL), dimethyl 4-cyano-4-[3-(cyclopentyloxy)-4-methoxyphenyl]-1,1-cyclohexanedicarboxylate (3.5 g, 8.43 mmol, 1.0 equivalent), water (0.455 g, 25.3 mmol, 3.0 equivalents), pyridine (0.66 g, 8.43 mmol, 1.0 equivalent), and lithium chloride (1.07 g, 25.3 mmol, 3.0 equivalents). The solution was stirred at 130 °C for 6.5 hours. The reaction solution was then cooled and diluted with 1% HCl and *t*-butyl methyl ether. The layers were separated and the organic layer was washed with water twice and with brine once. The solution was concentrated to a clear oil. Water was removed by reconcentrating once with methanol. The product was obtained in quantitative yield as a clear oil and as a mixture of *cis* and *trans* isomers in about a 1:1 ratio.

Mass spectrometry gave *m/z* = 372 (M+H⁺)⁺

(¹H NMR, 300 MHz, CDCl₃, δ ppm) δ 1.29 (t, 3H, ethyl CH₃), δ 1.55-1.70 (m, 2H, ring CH₂), δ 1.75-2.30 (m, 14H, ring CH₂'s), δ 2.75-2.80 (m, 1H, CHCO₂Et), 3.85 (s, 3H, OMe), δ 4.13-4.22 (q, 2H, ethyl CH₂), δ 4.8 (m, 1H, ring CH), δ 6.8-7.0 (m, 3H, aromatic)

Example 8**Preparation of 4-cyano-4-(3-cyclopentyloxy-4-methoxyphenyl)-cyclohexanecarboxylic acid**

The isomeric mixture (approximately 1 to 1 ratio) of the methyl esters prepared in Example 5 (2.94 g, 8.2 mmol, 1.0 equivalent) was dissolved in *n*-butanol (30 mL) under a nitrogen atmosphere. Potassium *n*-butoxide (1.8 g, 16.5 mmol, 2.0 equivalent) was added and the mixture was stirred 6-18 hours to give a ratio of *cis* to *trans* isomers of 14 to 1.

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The same procedure was used to treat the ethyl esters and gave a ratio of 8 to 1. The ratios were monitored using HPLC (15 cm Supelcocal LC-ABZ, 40/60/0.1 [acetonitrile/water/TFA], 1.5 mL/min., 215 nm UV)

- 5 To hydrolyze the equilibrated ester product, two drops of water were added to the reaction solution and the solution was stirred until no ester could be detected. The reaction was then diluted with *t*-butyl methyl ether and 5% HCl (the pH of the aqueous layer was between 1-2). The layers were separated and the organic layer was washed with brine. The ratio of *cis* to *trans* acid was improved even further (to 121 to 1) by crystallizing the *cis/trans* mixture from 20 ml of hexanes/ethyl acetate (3/1)

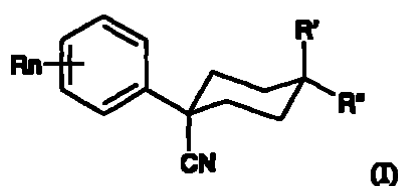
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What is claimed is:

- 1 A process for preparing a compound of formula (I)



where

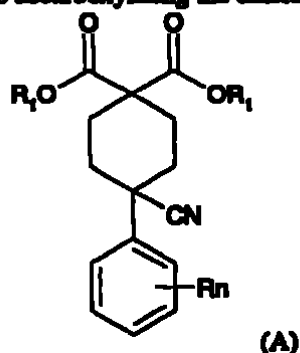
R is halo, C₁₋₆alkyl, C₁₋₆alkyl substituted with 1 to 4 halogens, C₁₋₆alkoxy, C₁₋₆alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons,

n is 1-5,

m is 0 - 6; and

R' and R'' are independently hydrogen or CO(O)X where X is hydrogen or C₁₋₆alkyl,

which process comprises decarboxylating the diacid or diester of Formula (A)



where R₁ is hydrogen or C₁₋₆ alkyl-ester forming group of 1-6 carbon atoms and R and n are the same as for Formula (I)

2. The process of claim 1 wherein the diester or diacid is combined with about 1 equivalent of a base, about 3 equivalents of water and about 3 equivalents of an alkali salt in a suitable solvent and heated to between about 100 to 150 °C for about 4-8 hours

3. The process of claim 1 wherein R₁ is hydrogen, methyl or ethyl and the base is pyridine and the salt is lithium chloride

4. The process of any one of claims 1 - 3 wherein n in R_n is 2 and one group is substituted on at the 3 position and the other group is substituted on the 4 position.

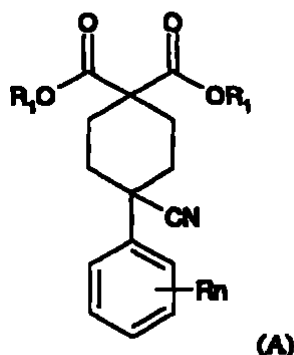
5. The process of any one of claims 1-4 wherein R₁ is methyl, one of R_n is methoxy, -O-CF₃, -O-CHF₂, or -O-CH₂CHF₂ and the other is C₄₋₆cycloalkyloxy

6. The process of any one of claim 1-6 wherein n in R_n is 2 and one is 3-cyclopentyloxy and a second R_n group is 4-methoxy

- 7 A compound of formula (A)

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wherein

R is halo, C₁₋₆alkyl, C₁₋₆alkyl substituted with 1 to 4 halogens, C₁₋₆alkoxy, C₁₋₆alkenyl, -O-(CH₂)_ncycloalkyl of 3-6 carbons,

n is 1-5,

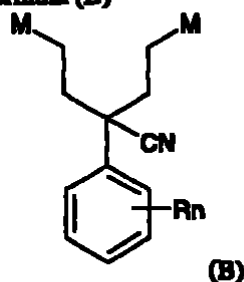
m is 0-6;

R₁ is hydrogen or a C₁₋₆ alkyl-ester forming group of 1-6 carbon atoms.

8 A compound according to claim 8 wherein n in R_n is 2 and R_n is methoxy, -O-CF₃, -O-CHF₂, or -O-CH₂CHF₂ and the other is C₄₋₆cycloalkyloxy

9 A compound according to any one of claims 8 or 9 wherein n in R_n is 2 and one is 3-cyclopentyloxy and a second R_n group is 4-methoxy.

10. A compound of Formula (B)



wherein

R is halo, C₁₋₆alkyl, C₁₋₆alkyl substituted with 1 to 4 halogens, C₁₋₆alkoxy, C₁₋₆alkenyl, -O-(CH₂)_ncycloalkyl of 3-6 carbons,

n is 1-5,

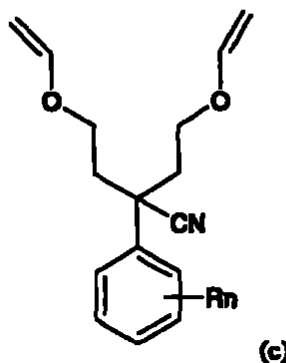
m is 0-6; and

M is OH, an activated hydroxyl group, or halo

12. A compound according to claim 11 wherein n in R_n is 2 and R_n is methoxy, -O-CF₃, -O-CHF₂, or -O-CH₂CHF₂ and the other is C₄₋₆cycloalkyloxy

13 A compound according to claim 11 or 12 wherein n in R_n is 2 and one is 3-cyclopentyloxy and the second R_n group is 4-methoxy

14. A compound of Formula (C)

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(c)

wherein

R is halo, C₁-alkyl, C₁-alkyl substituted with 1 to 4 halogens, C₁-alkoxy, C₁-alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons,

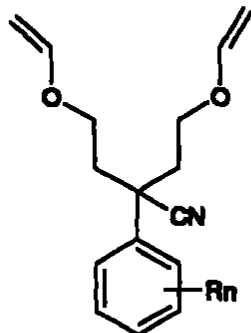
n is 1-5, and

m is 0 - 6.

15 A compound according to claim 14 wherein n in R_n is 2 and R_n is methoxy, -O-CF₃, -O-CHF₂, or -O-CH₂CHF₂ and the other is C₄-cycloalkyloxy

16. A compound according to claim 14 or 15 wherein n in R_n is 2 and one is 3-cyclopentyloxy and a second R_n group is 4-methoxy

17 A process for preparing a compound of Formula (C)



(C)

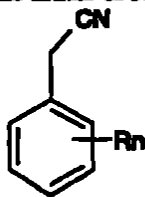
wherein

R is halo, C₁-alkyl, C₁-alkyl substituted with 1 to 4 halogens, C₁-alkoxy, C₁-alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons,

n is 1-5, and

m is 0 - 6

which comprises by treating the nitrile of formula (D)



(D)

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with 2-chloroethyl vinyl ether and a strong base

where, in Formula (D):

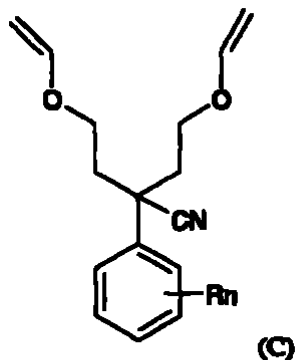
R is halo, C₁-alkyl, C₁-alkyl substituted with 1 to 4 halogens, C₁-alkoxy, C₁-alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons;

n is 1-5, and

m is 0 - 6

18 A process for preparing a compound of Formula (I) according to claim 1, which process comprises

a. converting the vinyl ether of Formula (C)

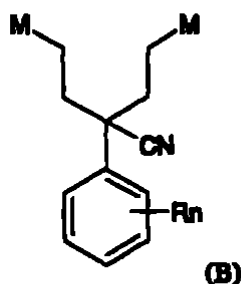


wherein R and n are halo, C₁-alkyl, C₁-alkyl substituted with 1 to 4 halogens, C₁-alkoxy, C₁-alkenyl, -O-(CH₂)_mcycloalkyl of 3-6 carbons,

n is 1-5,

m is 0 - 6,

to a compound of Formula (B)



where M is OH,

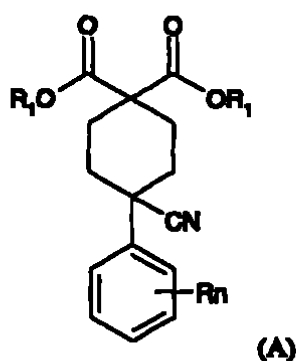
b. converting the hydroxyl group of Formula (B) to a compound of Formula (B)

where M is a tosylate, mesylate or a triflate,

c. converting the tosylate, mesylate or triflate in Formula (B) to a compound of Formula (B) where M is halo,

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d. treating the di-halo compound with dialkyl malonate to obtain a compound of Formula (A)



where R_1 is lower alkyl,

e. optionally saponifying the diester of Formula (A) to obtain a compound of Formula (A) where R_1 is hydrogen, and

f. decarboxylating a compound of Formula (A) where R_1 is hydrogen or C_{1-6} alkyl to obtain a compound for Formula (I) where one of R' is hydrogen and the other is $CO(O)X$ where X is C_{1-6} alkyl or hydrogen

19 The process of claim 18 wherein n in R_n is 2 and R_n is methoxy, $-O-CF_3$, $-O-CHF_2$, or $-O-CH_2CHF_2$ and the other is C_{4-6} cycloalkyloxy, M is tosylate, and thereafter iodo, and R_1 is methyl or ethyl.

20 A compound according to claim 18 or 19 wherein n in R_n is 2 and one is 3-cyclopentyloxy and the second is 4-methoxy

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ABSTRACT

This invention relates to a method for preparing 4-substituted-4-cyanocyclohexane carboxylates by forming the cyclohexane ring by treating a α,α -bis(2-haloethyl)-4-benzeneacetonitrile with a dialkyl malonate and decarboxylating the resulting diester

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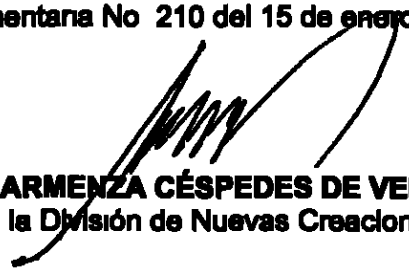
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	Evento	001
	Actuación	430
	Oficio No	1323
	Folios	001

Como resultado del examen de forma, realizado con fundamento en el artículo 38 de la Decisión 486 de la Comisión de la Comunidad Andina, se le comunica que

- El interesado debe informar la dirección del solicitante, debido a que en el petitorio no aparece la misma (Art 27-b, D 486/00)
- Debe allegar copia del documento en que consta la cesión del derecho a la patente otorgado por el inventor al solicitante o a su causante (art. 28-k, Dec 486/2000)

En cumplimiento del artículo 39 de la Decisión 486 de la Comisión de la Comunidad Andina, se le requiere para que dentro del término de dos (2) meses siguientes a la fecha de notificación del presente oficio, complete los requisitos anteriormente enunciados. En caso de no dar cumplimiento al presente requerimiento, la solicitud se considerará abandonada y perderá su prelación.

Notifíquese el presente oficio conforme a lo dispuesto en el artículo 8 de la Resolución Reglamentaria No 210 del 15 de enero del 2001


ALIX CARMENZA CÉSPEDES DE VERGEL
Jefe de la División de Nuevas Creaciones

03 MAYO 2001

El anterior requerimiento fue notificado por fijación en lista de fecha _____

202027/202022

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