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**INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE
SYNTHESIS GAS FROM HYDROCARBON FEEDS**

FEDERALLY SPONSORED RESEARCH OR DEVELOPMENT

[0001] This invention was made with Government support under Grant No. DE-AR0000506 awarded by the U.S. Department of Energy. The government has certain rights in the invention.

TECHNICAL FIELD

[0002] The present invention generally relates to the production of synthesis gas, particularly using an internal combustion engine as a synthesis gas generator.

BACKGROUND

[0003] Many processes and operations produce gaseous streams of light hydrocarbons. Often these gaseous streams are at low pressure and may further comprise a variety of contaminants. Therefore, the gaseous streams may have little intrinsic value and the cost of removing contaminants and compressing the gas (such as to increase the pressure to allow introduction into a natural gas transmission pipeline) may be prohibitively expensive. Given these constraints, the gaseous streams are often disposed of by flaring, incinerating or venting.

[0004] Recently, there has been increased interest in making more productive use of these low quality hydrocarbon streams. One such area is using the hydrocarbon stream as a feed gas in synthesis gas (syngas) production. Syngas may be produced from partial combustion of organic feedstocks (light hydrocarbons, coal, petcoke, biomass, oil) and consists primarily of hydrogen (H₂) and carbon monoxide (CO). Syngas often contains contaminants (including H₂S, COS) depending on the starting raw material. Many syngas production processes utilize catalyst-based reformers to partially oxidize the organic feedstocks. Although catalysts are useful for increasing reaction rate and reducing reaction temperature, many catalysts are made from expensive materials, are subject to poisoning from sulfur compounds present in the gas stream, and clogging from soot and other particles.

[0005] Syngas is the starting material for production of a variety of chemicals. Syngas can also be used for power production in a gas turbine or an engine-based generator. Syngas can also be used to produce H₂, by converting the CO and water vapor to H₂ and carbon dioxide (CO₂) via the water-

gas-shift (WGS) process. The H₂ to CO ratio of the process gas typically needs to be carefully adjusted to meet the downstream applications demand.

[0006] Recently, US Pat. No. 9,169,773 (Bromberg et al.) disclosed a reformer-liquid fuel manufacturing system utilizing an engine to generate a hydrogen-rich gas. The systems disclosed operates at an air/fuel ratio, equivalence ratio, $2.5 < \phi < 4.0$. They also disclose for an effective engine-based reformer one may use homogenous charge compression ignition (HCCI), partial pre-mixed compression ignition (PCI) or reaction controlled compression ignition (RCCI). They report that “In flowing burner flames as well as in cylinder calculations, lower equivalence ratios result in higher energy released in the conversion, higher peak in cylinder temperatures, lower selectivity to hydrogen and CO...”

[0007] US Pat. No. 2,391,687 (Eastman et al.) discloses an engine for the generation of syngas which runs on 90% or greater pure O₂ and has an equivalence ratio of 2.8-4.0.

SUMMARY

[0008] To address the foregoing problems, in whole or in part, and/or other problems that may have been observed by persons skilled in the art, the present disclosure provides methods, processes, systems, apparatus, instruments, and/or devices, as described by way of example in implementations set forth below.

[0009] According to one embodiment, a method for using an internal combustion engine under fuel-rich conditions includes: starting the engine using a feed gas having an initial fuel-air equivalence ratio; increasing the fuel-air equivalence ratio incrementally to generate a fuel-rich feed gas; and while increasing the fuel-air equivalence ratio, adjusting one or more of a throttle, an ignition timing, a load coupled to the engine, a fuel pressure, power to a supercharger acting on the feed gas, and power to a preheater acting on the feed gas to maintain a fuel-air equivalence ratio of about 1.6 to 2.4. In one embodiment, an engine speed between about 1000 to 2000 rotations per minute (RPM) and a temperature of an exhaust gas less than about 900°C is maintained.

[0010] According to another embodiment, a method for operating an internal combustion engine under fuel-rich conditions includes: maintaining an initial set of conditions after startup of the engine for an exhaust backpressure, an intake manifold pressure, an engine speed, an ignition timing, a fuel

gas fuel-air equivalence ratio, and a fuel gas inlet temperature; increasing the fuel gas inlet temperature while maintaining the fuel gas fuel-air equivalence ratio, and monitoring methane and oxygen content of an engine exhaust gas; and adjusting ignition timing in response to the monitored methane and oxygen content.

[0011] According to another embodiment, a gas reformer system is configured for performing any of the methods disclosed herein.

[0012] According to another embodiment, a gas reformer system includes: an internal combustion engine comprising a fuel gas inlet, an exhaust gas outlet, a plurality of cylinders, an ignition timing system, a throttle, a fuel gas preheater, and a supercharger, wherein the internal combustion engine is configured for operating with a fuel gas fuel-air equivalence ratio of between about 1.6 and 2.4.

[0013] Other devices, apparatus, systems, methods, features and advantages of the invention will be or will become apparent to one with skill in the art upon examination of the following figures and detailed description. It is intended that all such additional systems, methods, features and advantages be included within this description, be within the scope of the invention, and be protected by the accompanying claims.

BRIEF DESCRIPTION OF THE DRAWINGS

[0014] The invention can be better understood by referring to the following figures. The components in the figures are not necessarily to scale, emphasis instead being placed upon illustrating the principles of the invention. In the figures, like reference numerals designate corresponding parts throughout the different views.

[0015] Figure 1 is a cross-sectional schematic view of an exemplary cylinder of an internal combustion engine according to some embodiments.

[0016] Figure 2 is a schematic view of a system to produce synthesis gas using an internal combustion engine according to some embodiments.

[0017] Figure 3 is a schematic view of a system to produce synthesis gas using an internal combustion engine according to some embodiments.

[0018] Figure 4 is a schematic view of a system to produce synthesis gas using an internal combustion engine according to some embodiments.

[0019] Figure 5 is an exemplary flowchart of a method for starting an engine under fuel-rich conditions according to some embodiments.

[0020] Figure 6 is an exemplary flowchart of a method for operating an engine under fuel-rich conditions according to some embodiments.

[0021] Figure 7 is a graph of fuel-air intake temperature for various fuel-air equivalence ratios according to some embodiments.

[0022] Figure 8 is a graph of H₂ to CO ratios for various fuel-air equivalence ratios according to some embodiments.

[0023] Figure 9 is a graph of fractional conversion of natural gas for various fuel-air equivalence ratios according to some embodiments.

DETAILED DESCRIPTION

[0024] As used herein, the term “syngas” refers to synthesis gas. In the context of the present disclosure, syngas is a mixture of at least carbon monoxide (CO) and diatomic hydrogen gas (H₂). Depending on the embodiment, syngas may additionally include other components such as, for example, water, air, diatomic nitrogen gas (N₂), diatomic oxygen gas (O₂), carbon dioxide (CO₂), sulfur compounds (e.g., hydrogen sulfide (H₂S), carbonyl sulfide (COS), sulfur oxides (SO_x), etc.), nitrogen compounds (e.g., nitrogen oxides (NO_x), etc.), metal carbonyls, hydrocarbons (e.g., methane (CH₄)), ammonia (NH₃), chlorides (e.g., hydrogen chloride (HCl)), hydrogen cyanide (HCN), trace metals and metalloids (e.g., mercury (Hg), arsenic (As), selenium (Se), cadmium (Cd), etc.) and compounds thereof, particulate matter (PM), etc.

[0025] As used herein, the term “lower hydrocarbon” refers to hydrocarbons of a low molecular weight, including but not limited to methane, ethane, propane and butane.

[0026] As used herein, the term “load” may mean an electric heater, dynamometer, water bath, etc. By increasing or decreasing load, the engine may be run at constant speed as input conditions are varied. One may also vary the load to modify the temperature within the engine or the output component ratios.

[0027] As used herein, the term “natural gas” refers to a mixture of hydrocarbon (HC) gases consisting primarily of methane and lesser amounts of higher alkanes. Depending on the embodiment, natural gas may additionally include non-HC species such as one or more of those noted above, as

well as carbon disulfide (CS_2) and/or other disulfides, and mercaptans (thiols) such as methanethiol (CH_3SH) and ethanethiol ($\text{C}_2\text{H}_5\text{SH}$), and thiophenes such as thiophene ($\text{C}_4\text{H}_4\text{S}$) and other organosulfur compounds.

[0028] The present disclosure provides methods for utilizing an internal combustion engine as a syngas generator. Additionally, the process may be used in conjunction with methanol production, as well as other chemical production processes. The methods disclosed may make use of a wide variety of hydrocarbon sources dispersed throughout the world. For example, oil and natural gas production wells are located in many remote areas of the U.S., and each individual well, compressor, pneumatic device, and storage vessel at the well may produce a hydrocarbon emission stream. Due to the small volume, low pressure, and potential contaminants of these hydrocarbon streams, they are often flared or vented. Compressing and purifying these disparate streams for collection into a natural gas transmission pipeline may be prohibitively expensive. The methods of the present disclosure may utilize mass-produced internal combustion engines operated under specified conditions to partially oxidize these hydrocarbons and produce syngas which may have a higher value and make collection of the syngas economically feasible.

[0029] Internal combustion engines have been developed and utilized for decades to produce power typically either for the propulsion of vehicles, drive mechanical devices, or generate electricity. In each of these uses, the focus has been on the efficient and complete combustion of a fuel to maximize the power produced. Internal combustion engines however have characteristics of interest for other applications. Characteristics such as heat management from the coolant and radiator system for controlling the engine temperature, the ability to create high pressures in the cylinder, the function of short residence times in the cylinder, and the control of valve pressures can all be used in processes for chemical conversions where the engine would serve as a chemical reactor. In this application, the focus becomes on controlling the engine parameters to maximize the desired chemical conversion as opposed to power generation. In some instances, the chemical reaction may create power. Alternatively, power may be added by externally turning the shaft in other instances. Any chemical reaction operating on short residence time and taking advantage of the characteristics listed above could potentially be adapted to using the engine as a chemical reactor. Here we describe one such application for the conversion of light hydrocarbons to syngas.

[0030] FIG. 1 illustrates a cross-sectional schematic view of one cylinder 105 of an internal combustion engine 100 according to various embodiments. The engine 100 may further comprise a combustion chamber 110 within the cylinder 105. A piston 115 may be positioned within the cylinder 105 and move up and down in the cylinder 105, thereby defining a variable volume of the combustion chamber 110. The combustion chamber 110 is at its minimum volume when the piston 115 is at its highest position in the cylinder 105 (referred to as top dead center (TDC)), and at its maximum volume when the piston 115 is at its lowest position in the cylinder 105 (referred to as bottom dead center (BDC)). The piston 115 may be coupled to a crankshaft 120 by a connecting rod 125. The engine 100 may further comprise a head 130 coupled to the top of the cylinders 105. The head 130 may house one or more intake valves 135 and one or more exhaust valves 140. Each intake valve 135 may serve to open or close an inlet port 145, while each exhaust valve 140 may serve to open or close an exhaust port 150. The inlet port 145 may be in fluid communication with the combustion chamber 110 to allow a mixture of fuel and oxidizer (referred to as the charge) to flow into the combustion chamber 110. The exhaust port 150 may also be in fluid communication with the combustion chamber 110 to allow exhaust gases to flow out of the combustion chamber 110. The head 130 may also comprise one or more spark plugs 155 that may extend at least partially into the combustion chamber 110 and provide an ignition source for the charge.

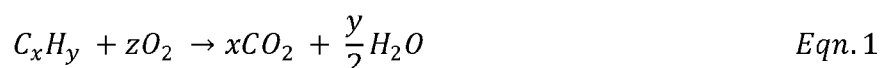
[0031] Although not shown in Fig. 1, the engine 100 may further comprise a preheater to increase a temperature of one or both of the fuel and oxidizer prior to entering the intake port 145, and a supercharger to increase a pressure of one or both of the fuel and oxidizer prior to entering the intake port 145. The preheater may comprise a heat exchanger to extract a portion of the heat, for example, of the exhaust gases from the engine 100. Alternatively, the preheater may obtain heat energy by other processes known in the art such as from electrically powered heaters, combustion of a secondary fuel, or scavenging heat from another process. The supercharger may comprise a turbine to extract energy from the exhaust gases from the engine 100. Alternatively, the supercharger may be powered by other processes known in the art such as an electric motor or scavenging energy from another process.

[0032] In certain non-limiting embodiments, the engine 100 may be adapted to operate using a 4-stroke process. The 4-stroke process may begin, for example, with the piston 115 at TDC and then starting to move downward. The intake valve 135 moves down, thereby placing the intake port 145 in fluid communication with the combustion chamber 110. The piston 115 moves downward allowing

the charge to enter the combustion chamber 110. As the piston 115 reaches BDC, the intake valve 135 closes. The piston 115 then moves upward, compressing the charge. As the piston 115 approaches TDC, the spark plug 155 generates a spark which ignites the charge. In various embodiments, as discussed further below, the spark may be generated prior to the piston 115 reaching TDC. As the charge burns, pressure within the combustion chamber 110 increases and forces the piston 115 downward. When the piston 115 reaches BDC, the exhaust valve 140 moves down, thereby placing the exhaust port 150 in fluid communication with the combustion chamber 110. As the piston 115 moves toward TDC, combustion gases from the burned charge are forced out of the exhaust port 150. When the piston 115 reaches TDC, the exhaust valve 140 closes and the cycle repeats.

[0033] Although the specifics of each process are not described herein, the engine 100 in various embodiments may be configured to operate according to a 2-stroke process, a 5-stroke process, a 6-stroke process, a compression ignition process (e.g., diesel), a jet engine, a turbine, a rotary engine, or any other engine type known in the art. The present disclosure focuses on a 4-stroke process because of its prevalence and ready availability, but no limitation on the scope of the disclosure should be inferred.

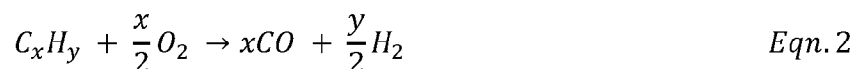
[0034] Typically, hydrocarbons and oxygen are the fuel and oxidizer, respectively, combusted in the engine 100. The general stoichiometric chemical equation for combustion of a hydrocarbon in oxygen is given by Equation 1:



Thus, for stoichiometric complete combustion, all of the hydrocarbon and oxygen react to form carbon dioxide and water. Thus, for stoichiometric complete combustion, a certain ratio of fuel to air is required. One measure of how the actual fuel-to-air ratio compares to the stoichiometric fuel-to-air ratio is the equivalence ratio (denoted as ϕ). The equivalence ratio is calculated by dividing the actual fuel-to-air ratio by the stoichiometric fuel-to-air ratio. Equivalence ratio values greater than 1 indicate fuel-rich conditions.

[0035] An engine 100 operated according to stoichiometric complete combustion is typically operated to extract useful work and heat. However, various embodiments may comprise operating the engine 100 at other than stoichiometric conditions in order to utilize the engine 100 as a chemical reactor. Certain embodiments may utilize fuel-rich conditions (i.e., less than the stoichiometric amount of oxygen to combust all of the hydrocarbon) to partially oxidize the hydrocarbon. Without

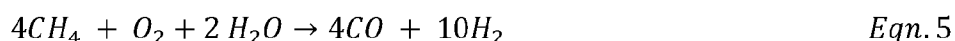
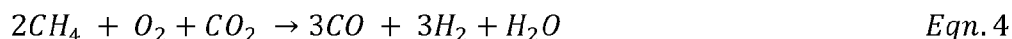
intending to be limited to a specific mechanism of action, under some conditions, the engine 100 may be operated as a reformer to produce synthesis gas (syngas) comprising hydrogen (H₂) and carbon monoxide (CO) according to the chemical reaction given by Equation 2:



If the hydrocarbon is methane (CH₄), then the partial oxidation reaction is given by Equation 3:



Additionally, other reactions are envisioned to possibly take place as well such as complete combustion (Equation 1), reforming reactions such as Equations 4 and 5 (shown for methane), and other such known reforming and combustion reactions.



[0036] FIG. 2 is a schematic diagram of a process 200 according to various embodiments in which the combustion engine 100 may be utilized as a chemical reactor to produce, for example, syngas. As discussed previously, the charge to the engine 100 may comprise an oxidizer and a fuel. The oxidizer may comprise air, enriched air, or a gas containing sufficient oxygen. The oxidizer may pass through a filter 205 to remove particulates and other solid contaminants, as well as liquid or gaseous contaminants such as water. The oxidizer flow may be controlled by a throttle 210. The throttle 210 may comprise any flow regulating device known in the art, and may be manually or electronically controlled. A pressure of the oxidizer may be increased by a supercharger 215. The supercharger 215 acts as a compressor to increase pressure, thereby allowing more oxygen to be delivered to each cylinder 105 of the engine 100. The supercharger 215 may be driven by an electric motor, or may be otherwise powered, such as by utilizing residual energy in the exhaust stream of the engine 100 or another process stream. The oxidizer may also pass through a heater 220 to increase the temperature of the oxidizer prior to entering the engine 100. The heater 220 may utilize electrically powered heating coils, or a heat exchanger that extracts energy from a process stream, such as the exhaust stream of the engine 100 (see, for example, FIG. 4). The fuel may also pass through a filter 225 to remove solid, liquid, or gaseous contaminants, and through a flow regulator 230. The temperature of the fuel stream may be increased by a heater 235 operated similarly to the oxidizer heater 220. A mixer 240 may mix the oxidizer and fuel in a desired ratio to form the charge delivered to each cylinder 105 of the engine 100. The engine 100 may then partially oxidize the charge as described

previously, producing syngas in the exhaust stream. Operation of the engine 100 may also produce mechanical power and heat.

[0037] In various embodiments, the process 200 may further comprise a central processing unit (not shown). The central processing unit may be in communication with and capable of activating and controlling one or more of the individual components of the process 200. The central processing unit may be capable of storing and executing computer code to initiate operation of the process 200 in response to monitored operating conditions of the engine 100 and analysis of the syngas produced by the engine 100. For example, the ratio of H₂ to CO in the exhaust stream of the engine 100 may be monitored, and the central processing unit may adjust one or more of the individual components of the process 200 in response to the monitored H₂ to CO ratio. Additionally, the central processing unit may adjust one or more of the individual components of the process 200 in response to monitored parameters of the engine 100, such as combustion temperature in one or more cylinders 105, inlet pressure, exhaust pressure, and the like.

[0038] FIG. 3 is a schematic diagram of a process 300 according to various embodiments in which a combustion engine 100 may be utilized as a chemical reactor to produce, for example, syngas. FIG. 3 illustrates that the engine 100 may be coupled to a generator 305 to convert mechanical power produced by the engine 100 to electrical power. The electrical power may be used to power the oxidizer heater 220, gas heater 235, or any other purpose.

[0039] FIG. 4 is a schematic diagram of a process 400 according to various embodiments in which a combustion engine 100 may be utilized as a chemical reactor to produce, for example, syngas. In process 400, residual heat in the syngas may be utilized to heat the oxidizer and the fuel. The oxidizer heater 220 and the fuel heater 235 may comprise heat exchangers in order to transfer heat from the syngas to the oxidizer and fuel streams. The heat exchangers may be any type known in the art, such as shell and tube, plate and shell, and plate fin and may operate in, for example, a parallel-flow, counter-flow, or cross-flow configuration.

[0040] Engines 100 operated under fuel-rich conditions may be susceptible to abnormal combustion known as detonation or knocking. Knocking occurs when a pocket of the charge ignites outside of a flame front created by the spark and can cause pressure within the cylinder 105 to rise beyond design limits. This increase in pressure has the potential to tear holes in the piston 115 or head 130, leading to catastrophic failure of the engine 100.

[0041] Various embodiments comprise methods to start up the engine 100 to reach operation at fuel-rich conditions and subsequently operate under steady-state conditions without (or with minimal) knocking. FIG. 5 illustrates a general flowchart of various embodiments of a method 500 for starting the engine 100 to reach operation at fuel-rich conditions. The engine 100 may be started using a feed gas having an initial fuel-air equivalence ratio at step 505. The fuel-air equivalence ratio may be increased incrementally at step 510 to generate a fuel-rich feed gas. At step 515, one or more of the following may be adjusted while increasing the fuel-air equivalence ratio to maintain an engine speed between about 1000 to 2000 RPM and a temperature of an exhaust gas less than about 900°C: the throttle 210, an ignition timing, a load coupled to the engine 100, a fuel pressure, power to the supercharger 215 acting on the feed gas, and power to the preheater 220, 235 acting on the feed gas.

[0042] Prior to starting the engine 100, initial operating conditions may be set such that a fuel pressure is a value typically lower than ambient pressure, the throttle 210 is partially opened to predetermined value, typically less than 50%, the ignition timing is set to a first predetermined value, and the load is coupled to the engine 100. In various embodiments, the ignition timing determines when a spark is generated by the spark plug 155 and is measured with respect to a rotational position of the crankshaft 120 when the piston 115 is at TDC. The ignition timing, in general, may be advanced to generate the spark prior to the piston 115 reaching TDC. Advancing the ignition timing allows combustion (or the desired amount of partial combustion) of the charge to be completed close to the point where the piston 115 reaches TDC. The ignition timing is typically expressed as the degrees of rotational movement of the crankshaft 120 prior to the piston 115 reaching TDC, or simply degrees before top dead center (BTDC). In various embodiments, the first predetermined ignition timing value may be approximately 8 degrees BDTC, or approximately 5 to approximately 15 degrees BDTC. In other embodiments, the first predetermined ignition timing value may be up to approximately 30 degrees BDTC.

[0043] After initiating starting the engine 100 and allowing the engine 100 to turn over for a short period of time until the engine fires, the fuel pressure may be increased typically to near ambient pressure to allow the engine 100 to run on its own. The engine speed should be monitored after starting. The ignition timing may be gradually advanced to a second predetermined value while adding load to maintain the engine speed between approximately 1000 to 2000 RPM. The second predetermined ignition timing value may be approximately 16 degrees BTDC. In other embodiments,

the second predetermined ignition timing value may range from approximately 8 degrees to approximately 28 degrees BTDC, preferably between approximately 10 and approximately 20 degrees BTDC.

[0044] As discussed previously, the supercharger 215 may be powered in certain embodiments by an electric motor. In such embodiments, the supercharger 215 may be initially powered, and then power may be incrementally increased while incrementally increasing the fuel pressure to maintain the engine speed between approximately 1000 to 2000 RPM. In various embodiments, the supercharger 215 may be initially powered at an initial predetermined value, typically a setting that supplies air similar to naturally aspirating the engine, and incrementally increased to a second predetermined value by increments that are approximately 10% to 15% of the range between those settings. In various embodiments, the fuel pressure may be incrementally increased by approximately 0.1 in. H₂O, though larger increments may be used.

[0045] The throttle 210 may be increased incrementally from the initial setting using increments that grow from 1% up to 10% until a final throttle position approximately 90% open is reached. While increasing the throttle 210, the fuel pressure may be increased in order to maintain the engine speed between approximately 1000 to 2000 RPM. If the performance of the engine 100 drops rapidly while adding fuel pressure, then the fuel-air mixture may be too rich. In this situation, fuel pressure may be decreased and load used to control engine speed. A temperature of the exhaust gases may also be monitored and should be maintained at less than 900°C by modifying one or more of the fuel pressure, the throttle 210, and the engine load.

[0046] The preheaters 220, 235 may be initially powered such that the preheaters 220, 235 maintain the temperature of the charge at approximately 200°C prior to entering the intake port 145. The ignition timing may then be advanced further BTDC. As is known in the art, the optimal ignition timing and specific process for advancing the ignition timing will depend on many factors including without limitation fuel conditions (temperature, pressure, presence of contaminants, etc.), timing of fuel injection, type and condition of ignition system, engine speed, engine load and the specific type of engine used. The power to the supercharger 215 may be incrementally increased to increase the air feed rate. As the supercharger power is increased and as the preheaters 220, 235 heat up, the engine speed may be maintained between approximately 1000 to 2000 RPM by adjusting the fuel pressure

and the load. Once a desired engine throughput is achieved, the supercharger power increases may be discontinued.

[0047] Once the preheaters 220, 235 reach approximately 200°C, the preheater temperature may be increased in predetermined increments typically no more than 15% of the setting to avoid overshooting the set temperature. As is known in the art, the use of automated control systems may allow for larger temperature increments without increasing the risk of overshooting the set temperature. The fuel pressure may be adjusted to maintain the engine speed between about 1000 to 2000 RPM while the preheater temperature increases. Once the desired fuel-air equivalence ratio is obtained, further increases in the preheater temperature may be discontinued. In various embodiments, the desired fuel-air equivalence ratio may be approximately 1.6 to 2.4.

[0048] If the engine 100 suddenly loses stability while increasing the preheater temperature, an operational temperature within the cylinder 105 may be too high for the present fuel-to-air ratio at the present manifold pressure. In this situation, the power to the preheaters 220, 235 may be turned off until stability is regained, and then the power to the preheaters 220, 235 may be turned back on, and ramping up of the preheater temperature and the fuel pressure may resume.

[0049] While air is often used to supply the oxidizer in the charge, enriched air (e.g., up to approximately 35 percent by volume O₂) may be used in various embodiments. The use of enriched air may increase engine throughput, reduce downstream costs per unit throughput, and improve liquid product collection and catalyst activity. Also humidified air or steam addition to the gas feed may be used. Increasing the humidity or adding additional steam increases the water vapor concentration in the cylinder enabling higher hydrogen yields through steam reforming reactions as described in Eqn. 5 above.

[0050] While methane is used here to describe the predominant fuel, a variety of hydrocarbon fuel compositions may be used in various embodiments. Operation of the system in many cases will utilize methane taken from natural gas pipelines, associated gas from oil wells, waste gas streams that would typically be flared, biogas streams, and other such gaseous light hydrocarbon streams. Pipeline natural gas is composed predominately of methane but will have levels of ethane between 1 to 6% and traces of other hydrocarbons, carbon dioxide, nitrogen, and other molecules. Likewise, associated gas from oil wells will have high concentrations of ethane and higher hydrocarbons, typically referred to as natural gas liquids, such as propane, butane, pentane, and hexane. In some instances, the natural

gas liquids are collected before the remaining natural gas is collected for the pipeline, utilized, or flared. Operation of the engine to produce syngas could be run with either this fuel with or without removal of the natural gas liquids by tailoring the operation parameters to maintain conversion and avoid soot production. Another fuel stream for the engine could be gas streams from various chemical, manufacturing, or industrial processes or storage systems either as wastes or by-products and such fuels can have a variety of light hydrocarbons. When these streams have a high enough fuel concentration (depending on the fuel composition) or can be treated to achieve such concentrations, it is envisioned that these would be potential fuel sources for syngas production to utilize these streams. Additionally, there are many sources of biogas that can be used as fuel for the production of syngas utilizing an engine. For the purposes of the present invention, biogas is defined as the gaseous stream produced from degradation of biomass materials containing predominantly methane and carbon dioxide with other trace components known in the biogas field. Examples of sources of biogas are landfills, sites for treatment of animal wastes, and waste water treatment plants. In the uses of biogas, pretreatment may be required to remove some carbon dioxide (depending on concentration) from the fuel to effectively operate the engine, though complete removal is not required for engine operation. Engine operation is generally tolerant to the presence of carbon dioxide. As described in the natural gas composition, ethane is typically found in natural gas streams. Other embodiments may use additional ethane either as a fuel or blended with natural gas as a fuel. The addition of hydrogen to the fuel stream may also provide benefit in some embodiments with various fuel compositions. This hydrogen could be obtained from outside sources or recycled from the engine operation or downstream processes. One embodiment would be to either recycle hydrogen selectively removed from the engine exhaust gas or recycle a fraction of the engine exhaust.

[0051] Once startup of the engine 100 to reach fuel-rich conditions is achieved (such as by implementing, for example, the method described above in reference to FIG. 5), FIG. 6 illustrates another general flow diagram of various embodiments of a method 600 for continued operation of the engine 100 under fuel-rich conditions. At step 605, an initial set of operating conditions may be maintained after startup of the engine 100 for an exhaust backpressure, an intake manifold pressure, the engine speed, the ignition timing, the fuel gas fuel-air equivalence ratio, and the fuel gas inlet temperature. The fuel gas inlet temperature may be increased at step 610 by increasing power to the preheater while maintaining the fuel-air equivalence ratio. A methane and oxygen content (assuming

methane is the hydrocarbon in the fuel) of the exhaust gas may be monitored. The ignition timing may be adjusted at step 615 in response to the monitored methane and oxygen content.

[0052] In various embodiments, the initial operating exhaust backpressure may be between approximately ambient to 5 bar absolute, the initial operating intake manifold pressure may be between approximately ambient to 2 bar absolute, the initial operating engine speed may be between approximately 1000-2000 RPM, the initial operating ignition timing may be between approximately 25 to 35 degrees BTDC, the initial operating fuel-air equivalence ratio may be between approximately 1.6 and 2.4, and the initial fuel gas temperature may be between approximately 200°C and 270°C. FIG. 7 illustrates approximate fuel gas temperature ranges for fuel-air equivalence ratios ranging from 1 to 2.

[0053] In various embodiments, adjusting the ignition timing in response to the monitored methane and oxygen comprises monitoring methane and oxygen slippage (i.e., unreacted methane and oxygen passing through the engine 100). If the methane content in the exhaust gas or the oxygen content in the exhaust gas exceed acceptable levels, the ignition timing may be advanced to reduce the slippage. The exhaust gas temperature may be monitored while advancing the ignition timing so that the exhaust gas temperature remains within the range specified in FIG. 7. In various embodiments, rather than advancing the ignition timing of all the cylinders 105 of the engine 100 equally, ignition timing may be adjusted individually for each cylinder 105 such that variability of the temperature of the exhaust gas of each individual cylinder 105 is within a range of approximately 75°C.

[0054] Under essentially steady-state operating conditions, FIG. 8 illustrates the expected ratio of H₂ to CO in the exhaust gas for a given fuel-air equivalence ratio. Thus, the engine 100 may be tuned to produce a desired H₂ to CO ratio as needed for downstream processes. FIG. 9 illustrates the fractional conversion of natural gas as the fuel for the engine 100 for a given fuel-air equivalence ratio. According to various embodiments, FIG. 8 may be used to determine the fuel-air equivalence ratio needed to produce a desired H₂ to CO ratio, and then FIG. 9 may be used to determine the expected fractional conversion of the fuel that may occur at the selected fuel-air equivalence ratio.

[0055] Example

[0056] An engine system was configured to produce syngas with natural gas sourced from the local utility natural gas pipeline. A commercially available 8-cylinder, 8.8L spark-ignited engine was

configured in a system to produce syngas with rich operation. Air was taken from the surrounding environment and a supercharger was used to boost the pressure to the inlet manifold pressure of near 2 bar. The natural gas was delivered from the utility pipeline meeting normal U.S. specifications for pipeline natural gas. The typical composition over the length of the runs described was 95 vol.% methane (CH_4), 4 vol.% ethane (C_2H_6), 1 vol.% carbon dioxide (CO_2) and unmeasured trace components. The air and natural gas mixtures were heated in excess of 200°C prior to mixing. The mixed feed was then fed to the engine cylinders through the intake manifold. With the aid of spark-ignition, the feed was converted to a syngas in the cylinder. The engine was operated at a speed of 1500 RPM and exhaust gas temperatures were maintained below 900°C . The produced syngas was collected through the exhaust manifold and maintained at a pressure between 4 to 5 bar by use of downstream pressure regulation.

[0057] Table 1 presents the syngas composition (average of four runs) from this example with the operation of the engine 100 according to the present disclosure.

[0058] Table 1. Syngas Composition

Syngas Component	Volume Percent (dry)
H_2	22.3
CO	14.5
O_2	0.15
CO_2	2.9
CH_4	1.3
N_2	58.9

[0059] The above Example is for illustrative purposes only and does not restrict the invention to the processes used in the example.

[0060] In general, terms such as “communicate” and “in . . . communication with” (for example, a first component “communicates with” or “is in communication with” a second component) are used herein to indicate a structural, functional, mechanical, electrical, signal, optical, magnetic, electromagnetic, ionic or fluidic relationship between two or more components or elements. As such, the fact that one component is said to communicate with a second component is not intended to

exclude the possibility that additional components may be present between, and/or operatively associated or engaged with, the first and second components.

[0061] It will be understood that various aspects or details of the invention may be changed without departing from the scope of the invention. Furthermore, the foregoing description is for the purpose of illustration only, and not for the purpose of limitation—the invention being defined by the claims.

CLAIMS

What is claimed is:

1. A method for using an internal combustion engine as a reactor under fuel-rich conditions, comprising:

starting the engine using a feed gas having an initial fuel-air equivalence ratio;
increasing the fuel-air equivalence ratio incrementally to generate a fuel-rich feed gas; and
while increasing the fuel-air equivalence ratio, adjusting one or more of a throttle, an ignition timing, a load coupled to the engine, a fuel pressure, power to a supercharger acting on the feed gas or part of the feed gas, and power to a preheater acting on the feed gas to maintain a fuel-air equivalence ratio to about 1.6 to 2.4.

2. The method of claim 1, wherein the feed gas comprises hydrocarbon compounds and an oxygen containing stream.

3. The method of claim 1, further comprising setting initial conditions prior to starting the engine as at least one of predetermined fuel pressure, partially open throttle, a first predetermined ignition timing value, and a load coupled to the engine.

4. The method of claim 3, wherein setting the partially open throttle comprises setting the throttle at a predetermined setting below 50%.

5. The method of claim 3, wherein setting the first predetermined ignition timing value comprises setting the ignition timing from about 5 to about 12 degrees before top dead center (BTDC).

6. The method of claim 1, wherein starting the engine comprises setting the fuel pressure to about 0 in. H₂O gauge and allowing the engine to start.
7. The method of claim 1, wherein an engine speed between about 1000 to 2000 rotations per minute (RPM) and a temperature of an exhaust gas less than about 900°C is maintained.
8. The method of claim 1, wherein adjusting the ignition timing comprises advancing the ignition timing to a second predetermined value while increasing the engine load to maintain the engine speed between about 1000 to 2000 RPM.
9. The method of claim 8, wherein the ignition timing second predetermined value is between about 8 degrees BTDC and about 28 degrees BTDC.
10. The method of claim 1, wherein adjusting the power to the supercharger comprises initially powering the supercharger.
11. The method of claim 10, further comprising increasing the power to the supercharger while increasing the fuel pressure to maintain the engine speed between about 1000 to 2000 RPM.
12. The method of claim 1, wherein adjusting the throttle comprises increasing the throttle while increasing the fuel pressure and the engine load to maintain the engine speed between about 1000 to 2000 RPM.

13. The method of claim 1, further comprising monitoring the exhaust gas temperature and modifying one or more of the fuel pressure, the throttle, and the engine load to maintain the exhaust gas temperature less than about 900°C.
14. The method of claim 1, wherein adjusting the power to the preheater comprises initially powering the preheater.
15. The method of claim 14, further comprising setting the preheater to an initial temperature, and raising the fuel pressure as the preheater temperature increases while maintaining the engine speed between about 1000 to 2000 RPM.
16. The method of claim 15, wherein the preheater initial temperature is about 200°C.
17. The method of claim 1, further comprising adjusting the ignition timing to a third predetermined value.
18. The method of claim 17, wherein the ignition timing third predetermined value is from about 20 to about 30 degrees BTDC.
19. The method of claim 11, further comprising increasing the power to the supercharger while adjusting the fuel pressure and the engine load to maintain the engine speed between about 1000 to 2000 RPM until a desired engine volumetric throughput is reached.

20. The method of claim 15, further comprising when the preheater reaches the initial temperature, increasing the preheater temperature while adjusting the fuel pressure to maintain the engine speed between about 1000 to 2000 RPM until the fuel-air equivalence reaches about 1.6 to 2.4.

21. The method of claim 1, wherein the fuel gas initial fuel-air equivalence ratio is about 1.

22. A method for operating an internal combustion engine under fuel-rich conditions, comprising:
maintaining a maintenance operating set of conditions after startup of the engine for an exhaust backpressure, an intake manifold pressure, an engine speed, an ignition timing, a fuel gas fuel-air equivalence ratio, and a fuel gas inlet temperature;

increasing the fuel gas inlet temperature while maintaining the fuel gas fuel-air equivalence ratio, and monitoring methane and oxygen content of an engine exhaust gas; and

adjusting ignition timing in response to the monitored methane and oxygen content.

23. The method of claim 22, wherein the initial operating exhaust backpressure is between ambient to 5 bar absolute.

24. The method of claim 22, wherein the operating intake manifold pressure is between about ambient to 2 bar absolute.

25. The method of claim 22, wherein the initial operating engine speed is between about 1000 to 2000 rotations per minute (RPM).
26. The method of claim 22, wherein the initial operating ignition timing is between about 25 to 35 degrees before top dead center (BTDC).
27. The method of claim 22, wherein the operating fuel gas fuel-air equivalence ratio is about 1.6 to 2.4.
28. The method of claim 22, wherein the initial operating fuel gas inlet temperature is between about 200°C and 270°C.
29. The method of claim 22, wherein adjusting the ignition timing in response to the monitored methane and oxygen content comprises advancing the ignition timing if the monitored methane or oxygen content increases beyond acceptable levels.
30. The method of claim 22, further comprising monitoring an exhaust gas temperature from each cylinder of the engine.
31. The method of claim 30, further comprising individually adjusting an ignition timing of each cylinder to reduce variability in the exhaust gas temperature of each cylinder to within a range of about 75°C.

32. The method of claim 22, wherein the exhaust gas comprises a combination of at least two of hydrogen (H₂), carbon monoxide (CO), nitrogen (N₂), water vapor (H₂O), carbon dioxide (CO₂), and trace components.

33. A gas reformer system configured for performing the method of any of the preceding claims.

34. The gas reformer system of claim 33, wherein the gas reformer system is configured to produce a syngas.

35. The gas reformer system of claim 34, wherein the syngas comprises hydrogen (H₂) and carbon monoxide (CO).

36. A gas reformer system, comprising:

an internal combustion engine comprising a fuel gas inlet, an exhaust gas outlet, a plurality of cylinders, an ignition timing system, a throttle, a fuel gas preheater, and a supercharger;

wherein the internal combustion engine is configured for operating with a fuel gas fuel-air equivalence ratio of between about 1.6 and 2.4.

37. The gas reformer system of claim 36, wherein the internal combustion engine is configured for individually adjusting a fuel gas fuel-air equivalence ratio, a fuel gas inlet temperature, an inlet manifold pressure, an ignition timing, an engine speed, an exhaust manifold pressure, and an exhaust gas temperature in order to operate with a fuel gas fuel-air equivalence ratio of between about 1.6 and 2.4.

38. The gas reformer system of claim 36, wherein the internal combustion engine is configured to produce an exhaust gas comprising hydrogen (H₂) and carbon monoxide (CO).

ABSTRACT

An internal combustion engine is operated at fuel-rich conditions by adjusting one or more operating parameters such as, for example, a throttle, an ignition timing, a load coupled to the engine, a fuel pressure, power to a supercharger, and power to a preheater to maintain a specified engine speed and a temperature of an exhaust gas. Operating the engine under these conditions allows the engine to function as a reformer producing a synthesis gas comprising hydrogen and carbon monoxide.

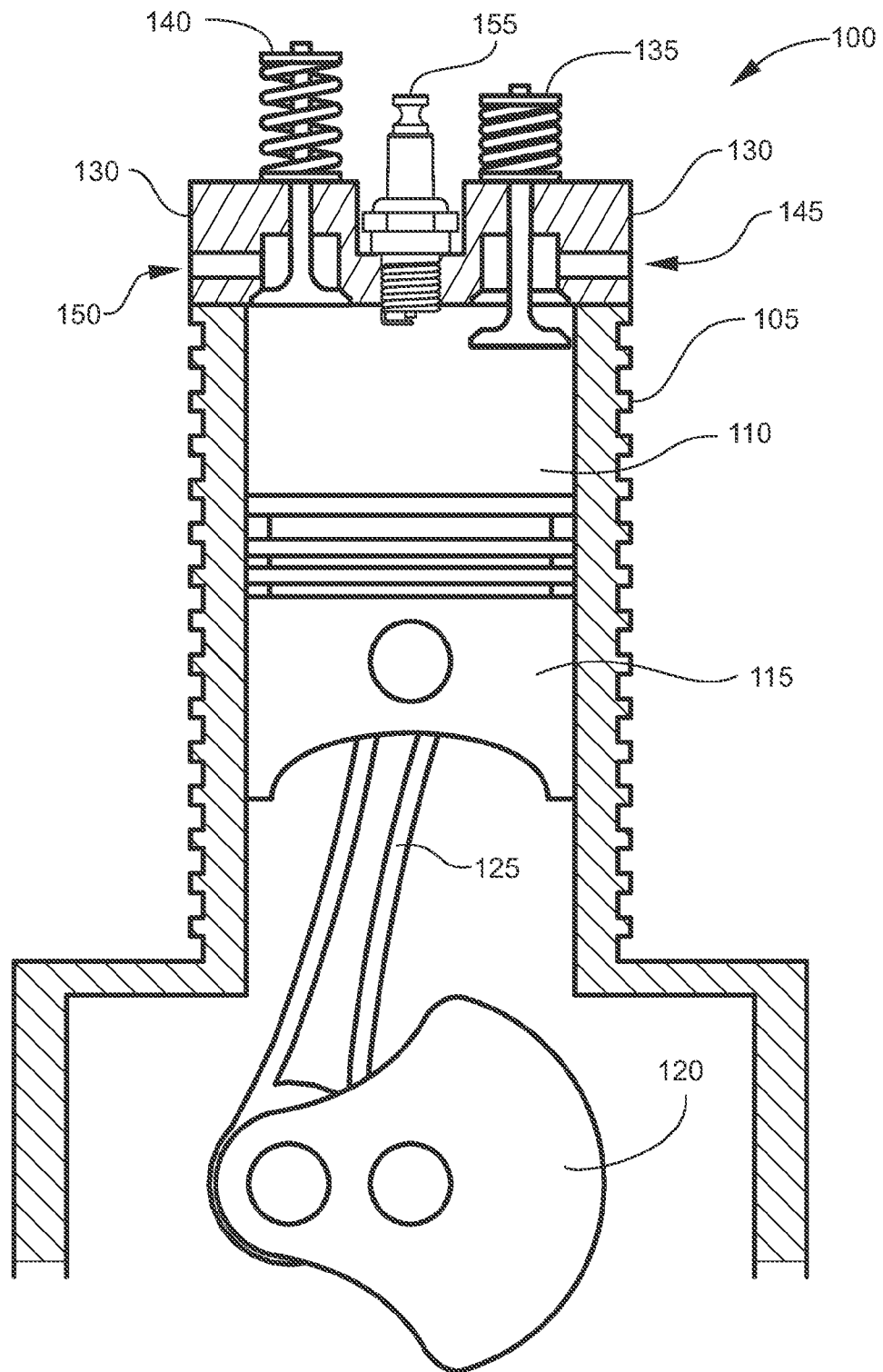


FIG. 1

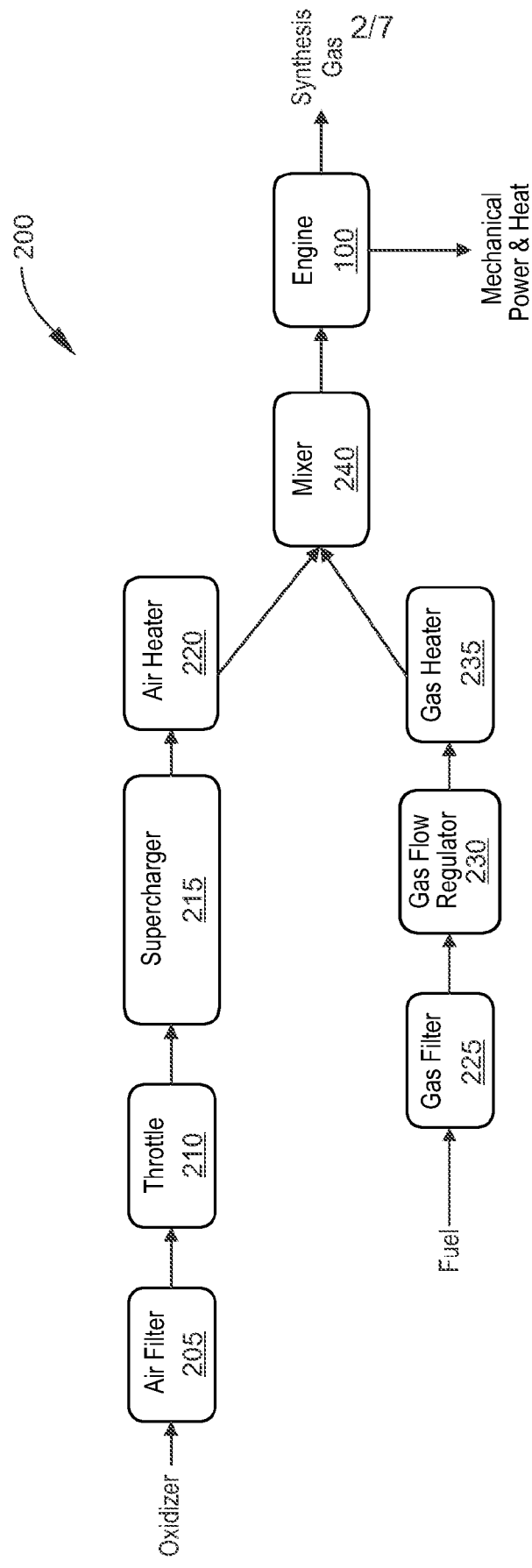


FIG. 2

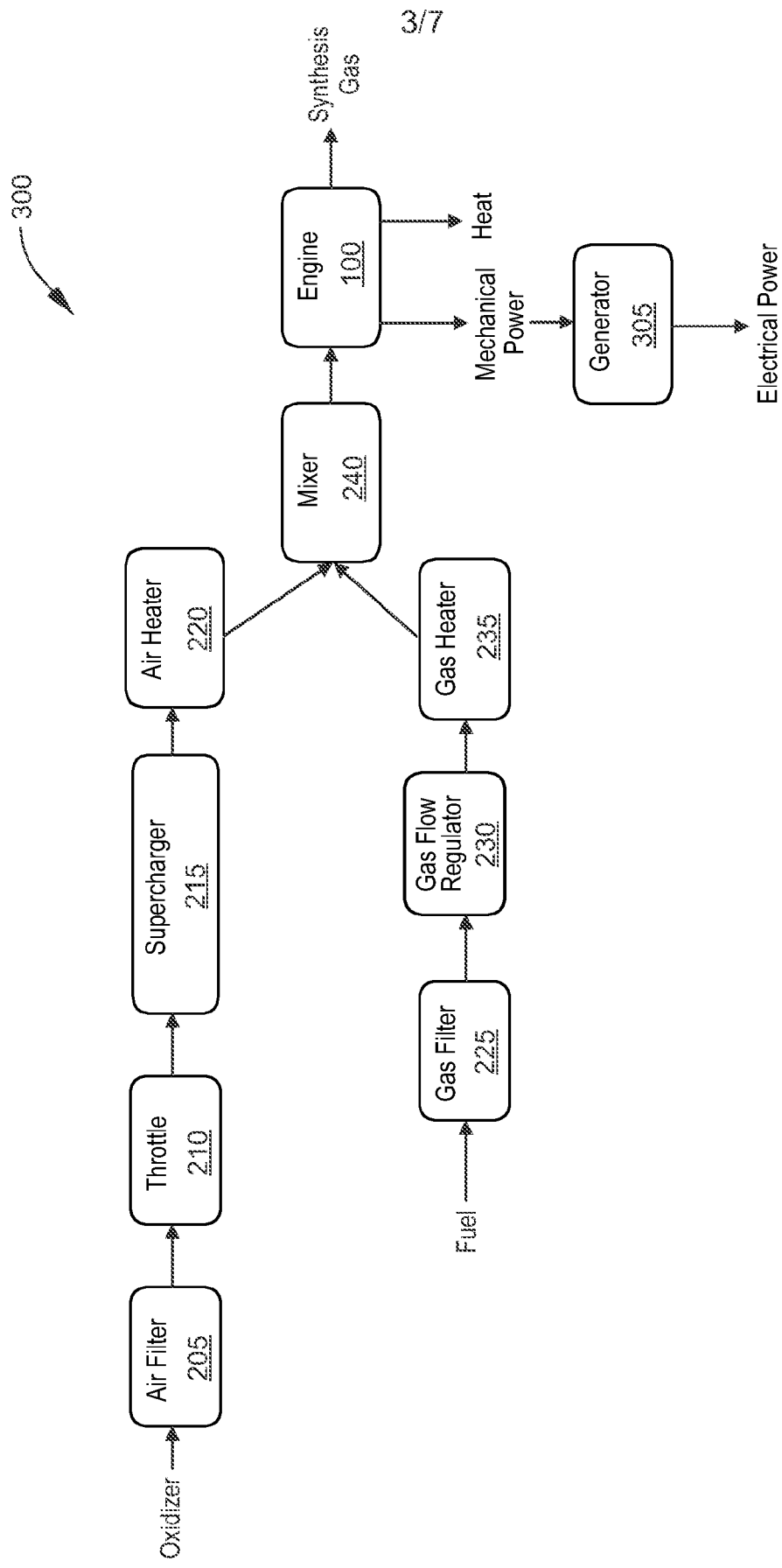


FIG. 3

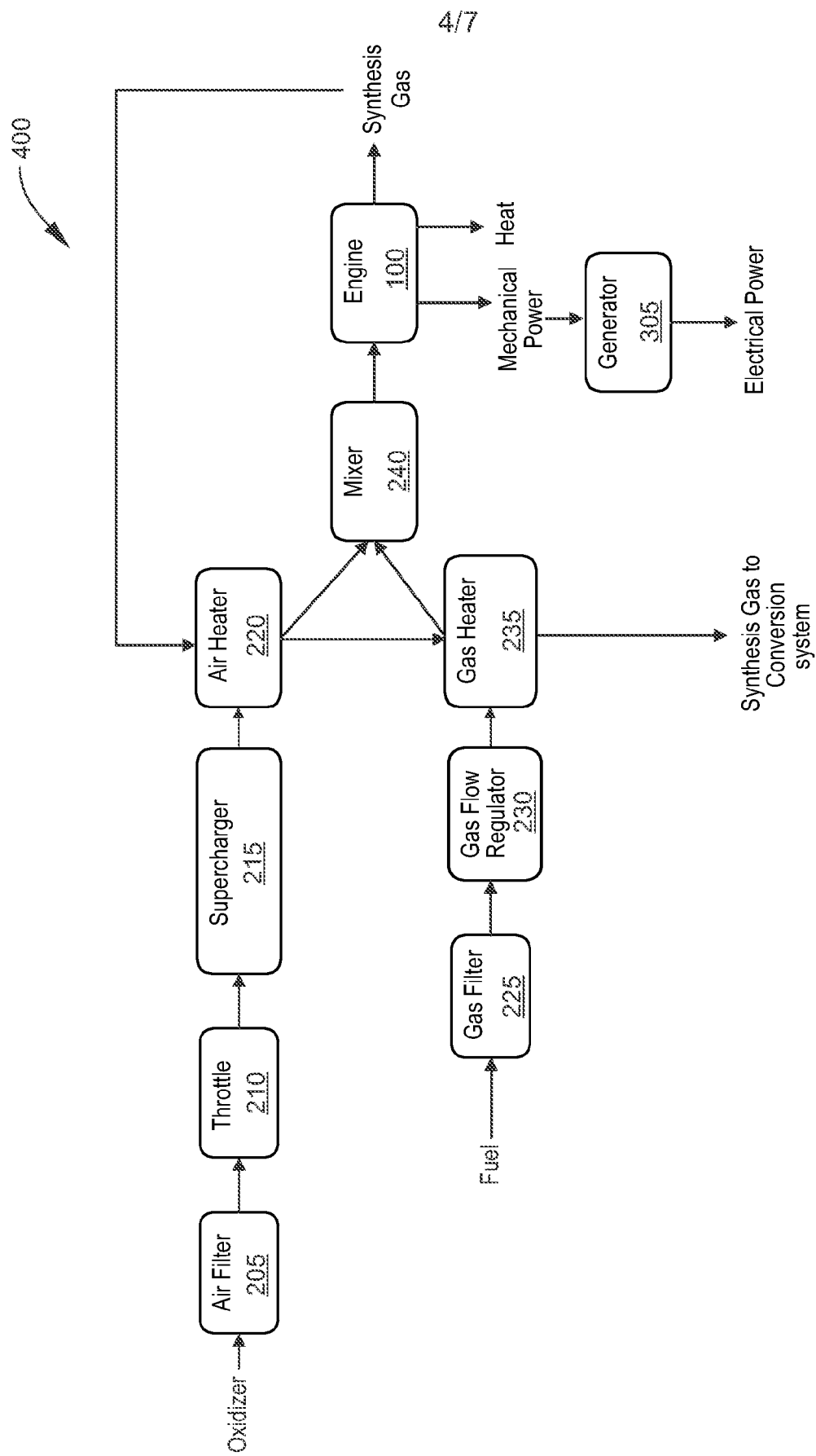


FIG. 4

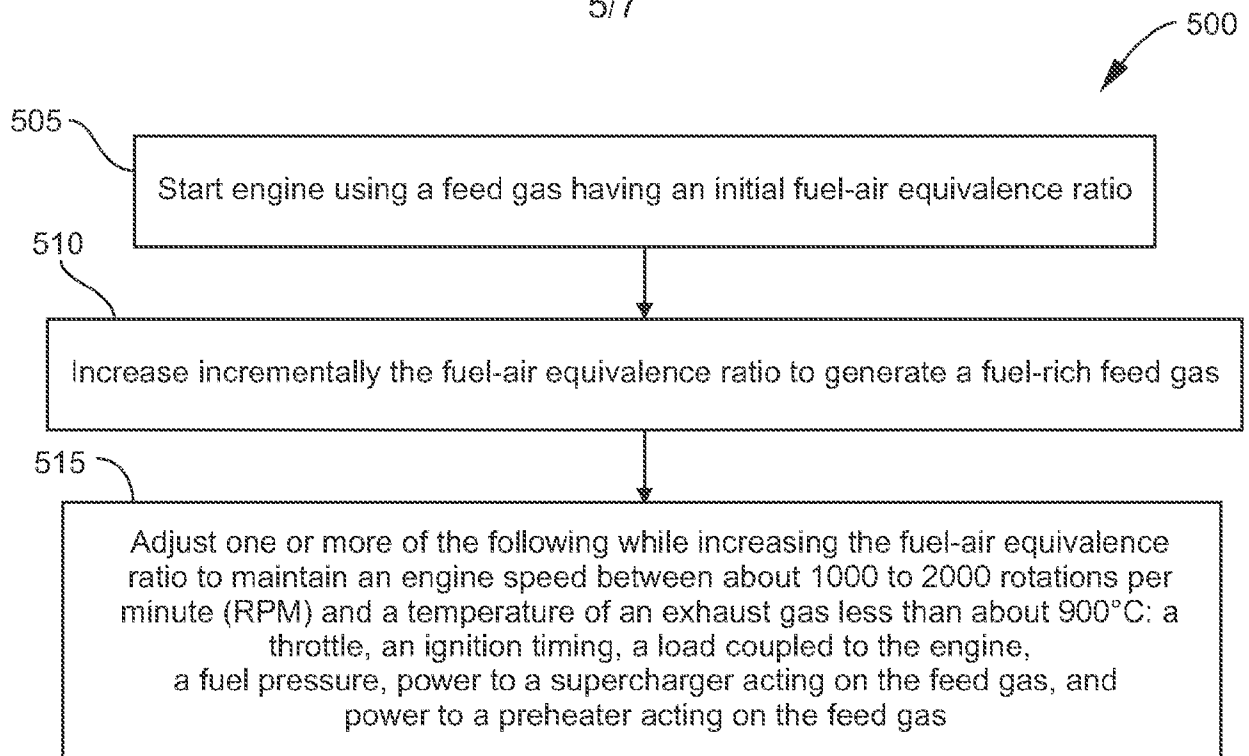


FIG. 5

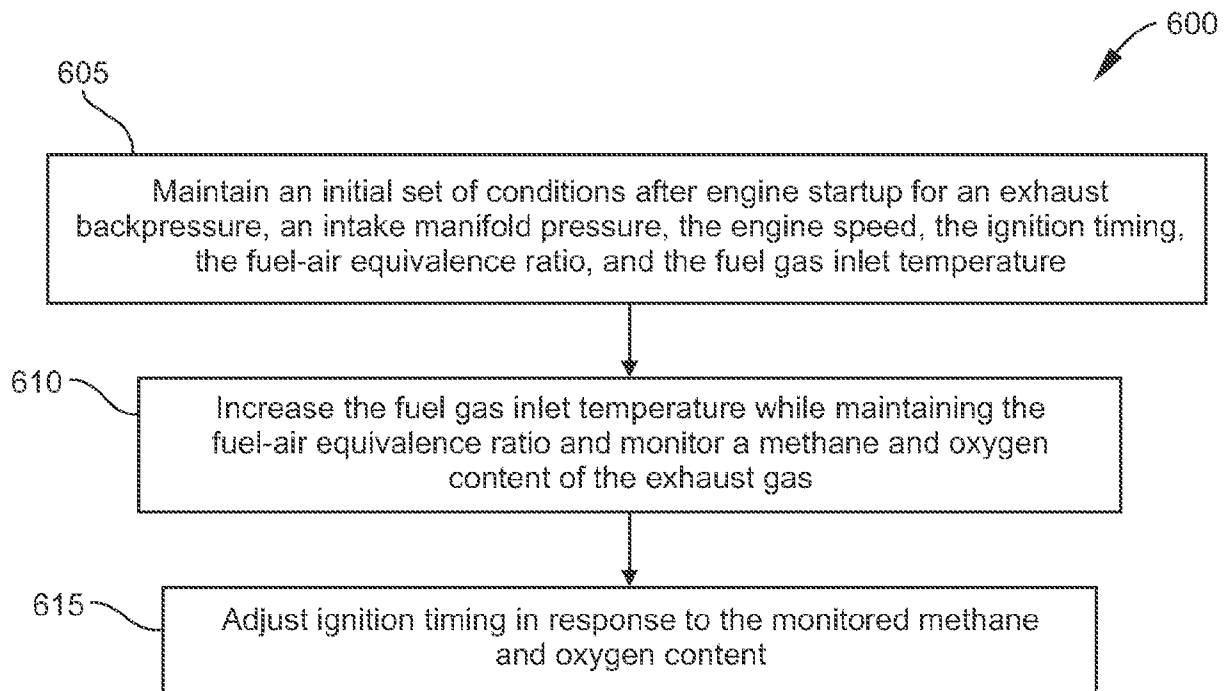


FIG. 6

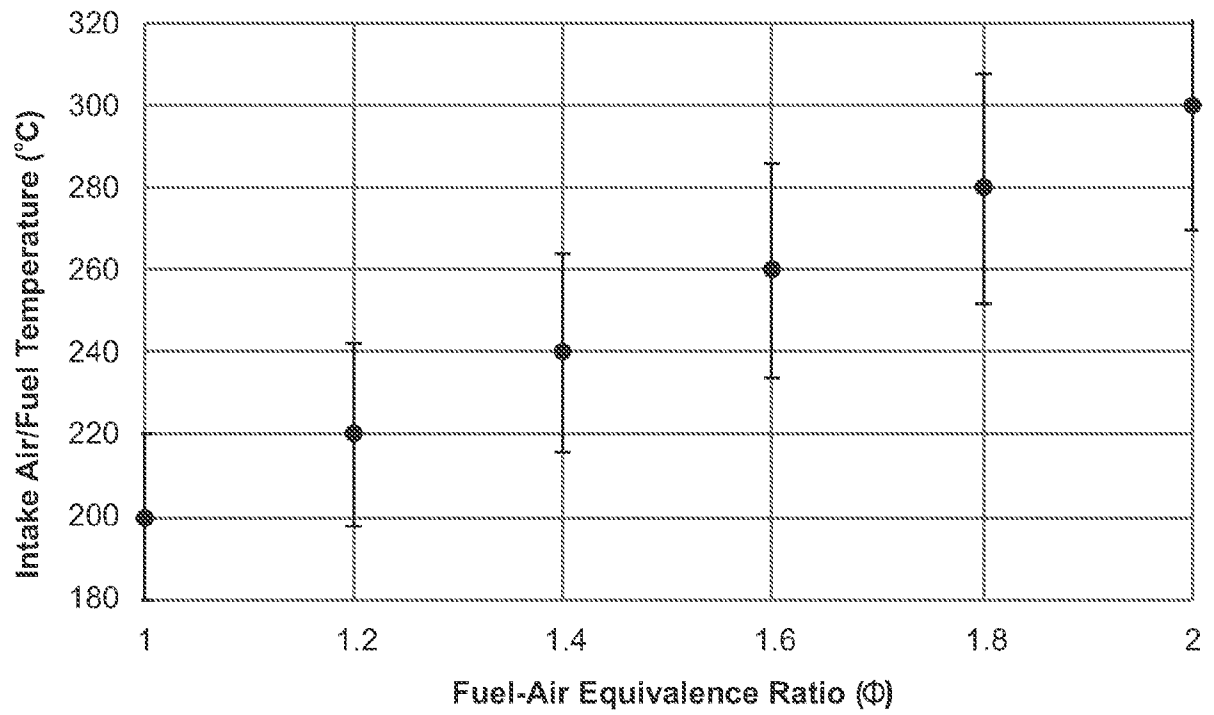


FIG. 7

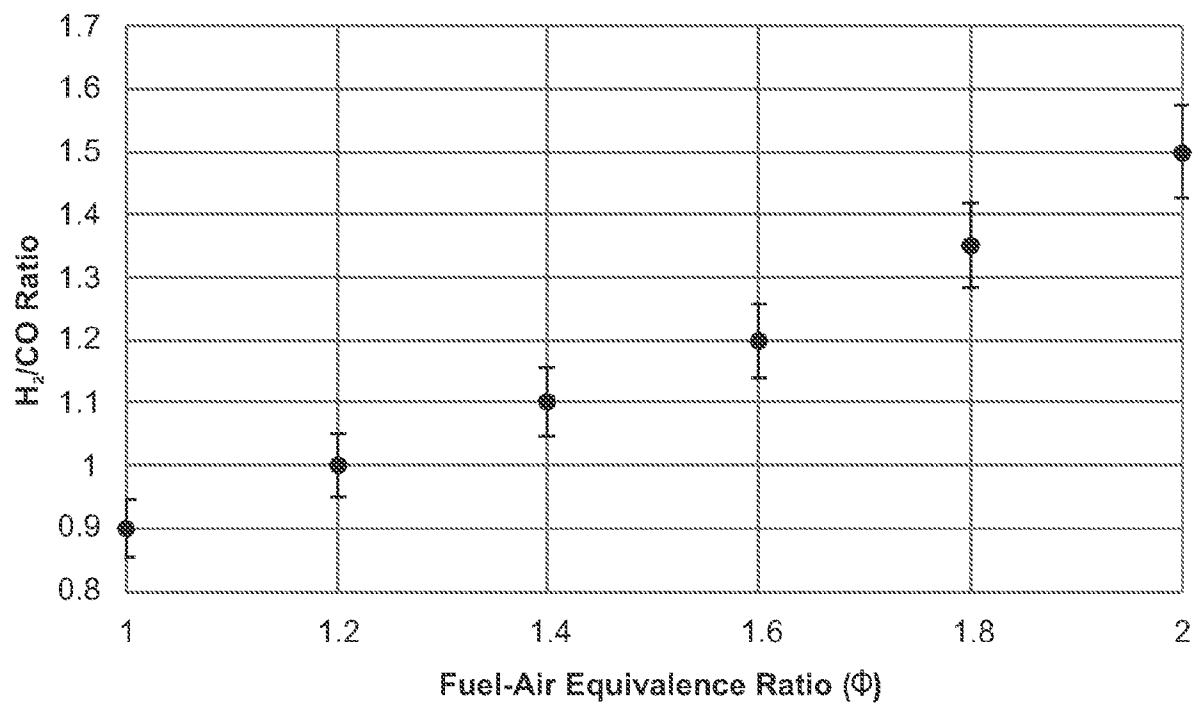


FIG. 8

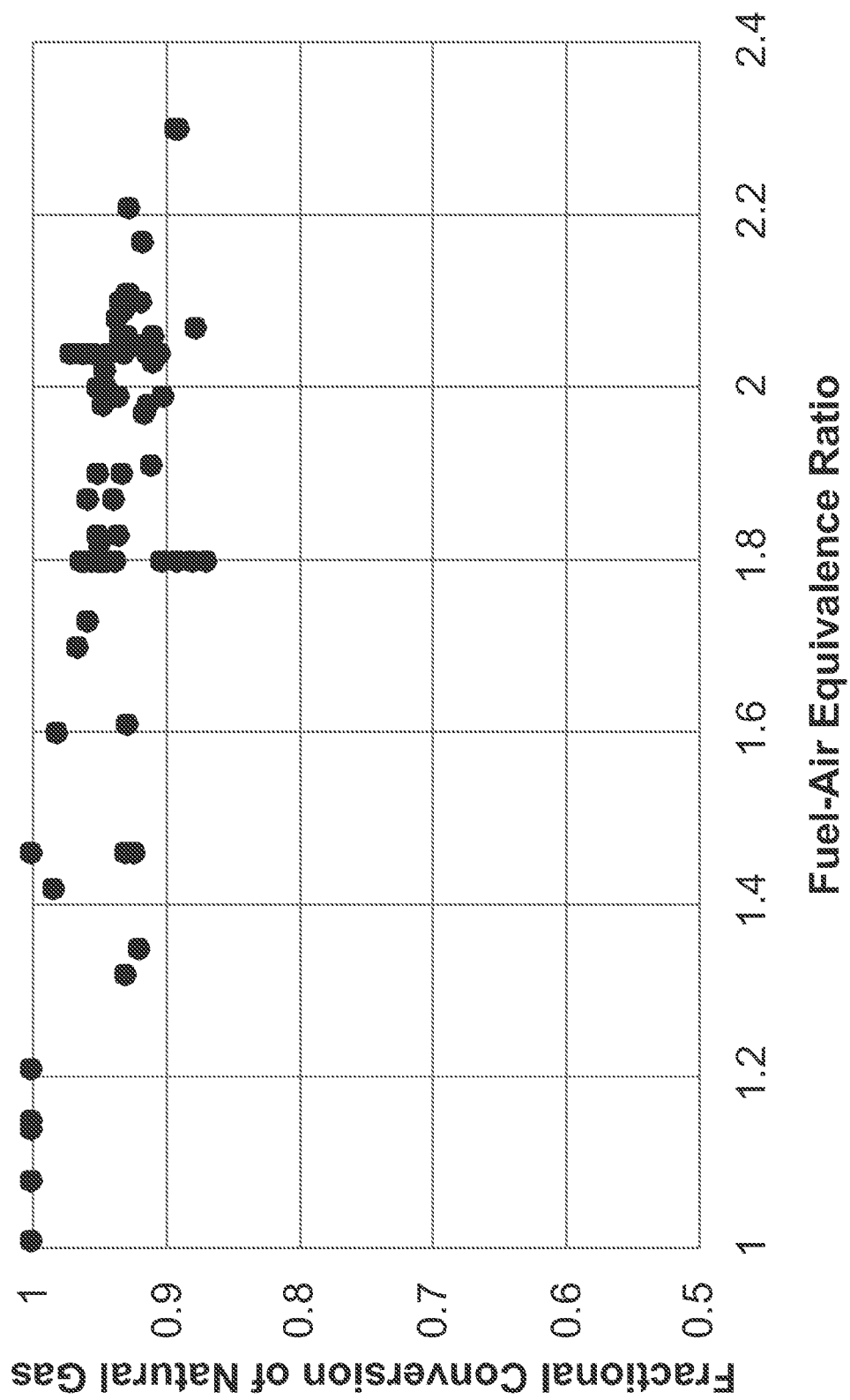


FIG. 9

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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	121-79-PROV
		Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		
The application data sheet is part of the provisional or nonprovisional application for which it is being submitted. The following form contains the bibliographic data arranged in a format specified by the United States Patent and Trademark Office as outlined in 37 CFR 1.76. This document may be completed electronically and submitted to the Office in electronic format using the Electronic Filing System (EFS) or the document may be printed and included in a paper filed application.			

Secrecy Order 37 CFR 5.2:

☐	Portions or all of the application associated with this Application Data Sheet may fall under a Secrecy Order pursuant to 37 CFR 5.2 (Paper filers only. Applications that fall under Secrecy Order may not be filed electronically.)
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Inventor	1				Remove	
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	John	Reeves	Carpenter	III		
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Address 2	3040 Cornwallis Road					
City	Research Triangle Park	State/Province	NC			
Postal Code	27709	Country i	US			
Inventor	2				Remove	
Legal Name						
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	David	Douglas	Barbee			
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	Apoorv		Agrawal			
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Application Data Sheet 37 CFR 1.76		Attorney Docket Number		121-79-PROV	
		Application Number			
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS				
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Address 2	3040 Cornwallis Road				
City	Research Triangle Park		State/Province	NC	
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All Inventors Must Be Listed - Additional Inventor Information blocks may be generated within this form by selecting the Add button.					Add

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Email Address	docket@olivelawgroup.com	Add Email	Remove Email

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Title of the Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		
Attorney Docket Number	121-79-PROV	Small Entity Status Claimed	☐
Application Type	Provisional		
Subject Matter	Utility		
Total Number of Drawing Sheets (if any)	7	Suggested Figure for Publication (if any)	

Filing By Reference:

Only complete this section when filing an application by reference under 35 U.S.C. 111(c) and 37 CFR 1.57(a). Do not complete this section if application papers including a specification and any drawings are being filed. Any domestic benefit or foreign priority information must be provided in the appropriate section(s) below (i.e., "Domestic Benefit/National Stage Information" and "Foreign Priority Information").

For the purposes of a filing date under 37 CFR 1.53(b), the description and any drawings of the present application are replaced by this reference to the previously filed application, subject to conditions and requirements of 37 CFR 1.57(a).

Application number of the previously filed application	Filing date (YYYY-MM-DD)	Intellectual Property Authority or Country ⁱ

Publication Information:

☐ Request Early Publication (Fee required at time of Request 37 CFR 1.219)
☐ Request Not to Publish. I hereby request that the attached application not be published under 35 U.S.C. 122(b) and certify that the invention disclosed in the attached application has not and will not be the subject of an application filed in another country, or under a multilateral international agreement, that requires publication at eighteen months after filing.

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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	121-79-PROV
		Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		

Representative Information:

Representative information should be provided for all practitioners having a power of attorney in the application. Providing this information in the Application Data Sheet does not constitute a power of attorney in the application (see 37 CFR 1.32). Either enter Customer Number or complete the Representative Name section below. If both sections are completed the customer Number will be used for the Representative Information during processing.

Please Select One:	☒ Customer Number	☐ US Patent Practitioner	☐ Limited Recognition (37 CFR 11.9)
Customer Number	79683		

Domestic Benefit/National Stage Information:

This section allows for the applicant to either claim benefit under 35 U.S.C. 119(e), 120, 121, 365(c), or 386(c) or indicate National Stage entry from a PCT application. Providing benefit claim information in the Application Data Sheet constitutes the specific reference required by 35 U.S.C. 119(e) or 120, and 37 CFR 1.78.

When referring to the current application, please leave the "Application Number" field blank.

Prior Application Status			Remove
Application Number	Continuity Type	Prior Application Number	Filing or 371(c) Date (YYYY-MM-DD)
Additional Domestic Benefit/National Stage Data may be generated within this form by selecting the Add button.			Add

Foreign Priority Information:

This section allows for the applicant to claim priority to a foreign application. Providing this information in the application data sheet constitutes the claim for priority as required by 35 U.S.C. 119(b) and 37 CFR 1.55. When priority is claimed to a foreign application that is eligible for retrieval under the priority document exchange program (PDX)ⁱ the information will be used by the Office to automatically attempt retrieval pursuant to 37 CFR 1.55(i)(1) and (2). Under the PDX program, applicant bears the ultimate responsibility for ensuring that a copy of the foreign application is received by the Office from the participating foreign intellectual property office, or a certified copy of the foreign priority application is filed, within the time period specified in 37 CFR 1.55(g)(1).

Application Number	Country ⁱ	Filing Date (YYYY-MM-DD)	Access Code ⁱ (if applicable)	Remove
Additional Foreign Priority Data may be generated within this form by selecting the Add button.				Add

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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	121-79-PROV
		Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications

This application (1) claims priority to or the benefit of an application filed before March 16, 2013 and (2) also contains, or contained at any time, a claim to a claimed invention that has an effective filing date on or after March 16, 2013. ☐ NOTE: By providing this statement under 37 CFR 1.55 or 1.78, this application, with a filing date on or after March 16, 2013, will be examined under the first inventor to file provisions of the AIA.	
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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	121-79-PROV
		Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		

Authorization or Opt-Out of Authorization to Permit Access:

When this Application Data Sheet is properly signed and filed with the application, applicant has provided written authority to permit a participating foreign intellectual property (IP) office access to the instant application-as-filed (see paragraph A in subsection 1 below) and the European Patent Office (EPO) access to any search results from the instant application (see paragraph B in subsection 1 below).

Should applicant choose not to provide an authorization identified in subsection 1 below, applicant **must opt-out** of the authorization by checking the corresponding box A or B or both in subsection 2 below.

NOTE: This section of the Application Data Sheet is **ONLY** reviewed and processed with the **INITIAL** filing of an application. After the initial filing of an application, an Application Data Sheet cannot be used to provide or rescind authorization for access by a foreign IP office(s). Instead, Form PTO/SB/39 or PTO/SB/69 must be used as appropriate.

1. Authorization to Permit Access by a Foreign Intellectual Property Office(s)

A. Priority Document Exchange (PDX) - Unless box A in subsection 2 (opt-out of authorization) is checked, the undersigned hereby **grants the USPTO authority** to provide the European Patent Office (EPO), the Japan Patent Office (JPO), the Korean Intellectual Property Office (KIPO), the State Intellectual Property Office of the People's Republic of China (SIPO), the World Intellectual Property Organization (WIPO), and any other foreign intellectual property office participating with the USPTO in a bilateral or multilateral priority document exchange agreement in which a foreign application claiming priority to the instant patent application is filed, access to: (1) the instant patent application-as-filed and its related bibliographic data, (2) any foreign or domestic application to which priority or benefit is claimed by the instant application and its related bibliographic data, and (3) the date of filing of this Authorization. See 37 CFR 1.14(h)(1).

B. Search Results from U.S. Application to EPO - Unless box B in subsection 2 (opt-out of authorization) is checked, the undersigned hereby **grants the USPTO authority** to provide the EPO access to the bibliographic data and search results from the instant patent application when a European patent application claiming priority to the instant patent application is filed. See 37 CFR 1.14(h)(2).

The applicant is reminded that the EPO's Rule 141(1) EPC (European Patent Convention) requires applicants to submit a copy of search results from the instant application without delay in a European patent application that claims priority to the instant application.

2. Opt-Out of Authorizations to Permit Access by a Foreign Intellectual Property Office(s)

☐ A. Applicant **DOES NOT** authorize the USPTO to permit a participating foreign IP office access to the instant application-as-filed. If this box is checked, the USPTO will not be providing a participating foreign IP office with any documents and information identified in subsection 1A above.

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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	121-79-PROV
		Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		

Applicant Information:

Providing assignment information in this section does not substitute for compliance with any requirement of part 3 of Title 37 of CFR to have an assignment recorded by the Office.

Applicant	1 Remove		
If the applicant is the inventor (or the remaining joint inventor or inventors under 37 CFR 1.45), this section should not be completed. The information to be provided in this section is the name and address of the legal representative who is the applicant under 37 CFR 1.43; or the name and address of the assignee, person to whom the inventor is under an obligation to assign the invention, or person who otherwise shows sufficient proprietary interest in the matter who is the applicant under 37 CFR 1.46. If the applicant is an applicant under 37 CFR 1.46 (assignee, person to whom the inventor is obligated to assign, or person who otherwise shows sufficient proprietary interest) together with one or more joint inventors, then the joint inventor or inventors who are also the applicant should be identified in this section. Clear			
☒ Assignee	Legal Representative under 35 U.S.C. 117		Joint Inventor
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If applicant is the legal representative, indicate the authority to file the patent application, the inventor is:			
Name of the Deceased or Legally Incapacitated Inventor:			
If the Applicant is an Organization check here. ☒			
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Mailing Address Information For Applicant:			
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Country	US	Postal Code	27709
Phone Number		Fax Number	
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Assignee Information including Non-Applicant Assignee Information:

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Application Data Sheet 37 CFR 1.76	Attorney Docket Number	121-79-PROV
	Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS	

Assignee	1
-----------------	---

Complete this section if assignee information, including non-applicant assignee information, is desired to be included on the patent application publication. An assignee-applicant identified in the "Applicant Information" section will appear on the patent application publication as an applicant. For an assignee-applicant, complete this section only if identification as an assignee is also desired on the patent application publication.

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If the Assignee or Non-Applicant Assignee is an Organization check here. ☐

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Phone Number		Fax Number		
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This Application Data Sheet **must** be signed by a patent practitioner if one or more of the applicants is a **juristic entity** (e.g., corporation or association). If the applicant is two or more joint inventors, this form must be signed by a patent practitioner, **all** joint inventors who are the applicant, or one or more joint inventor-applicants who have been given power of attorney (e.g., see USPTO Form PTO/AIA/81) on behalf of **all** joint inventor-applicants.

See 37 CFR 1.4(d) for the manner of making signatures and certifications.

Signature	/Nathan P. Letts/		Date (YYYY-MM-DD)	2017-09-29
First Name	Nathan	Last Name	Letts	Registration Number
				36,581

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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	121-79-PROV
		Application Number	
Title of Invention	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS		

This collection of information is required by 37 CFR 1.76. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 23 minutes to complete, including gathering, preparing, and submitting the completed application data sheet form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.**

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EFS ID:	30526530
Application Number:	62565844
International Application Number:	
Confirmation Number:	3652
Title of Invention:	INTERNAL COMBUSTION ENGINE AS A CHEMICAL REACTOR TO PRODUCE SYNTHESIS GAS FROM HYDROCARBON FEEDS
First Named Inventor/Applicant Name:	John Reeves Carpenter III
Customer Number:	79683
Filer:	Nathan P. Letts/Vishva Miyani
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File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1		121-79- PROV_2017-09-29_appn.pdf	142277	yes	24
			4f0d2677a57286471bc41298f377a3c161bf0e21		
	Multipart Description/PDF files in .zip description				
	Document Description		Start	End	
	Specification		1	16	
	Claims		17	23	
	Abstract		24	24	
Warnings:					
Information:					
2	Drawings-only black and white line drawings	121-79- PROV_2017-09-27_Figs1-9b.pdf	96196	no	7
			d09e093420aac51a345ea25844de05ecbd99b630		
Warnings:					
Information:					
3	Application Data Sheet	121-79-PROV_2017-09-29_ADS.pdf	1822961	no	9
			ee8ca27564d5653270059a86c19e999c84db6edc		
Warnings:					
Information:					
4	Fee Worksheet (SB06)	fee-info.pdf	29908	no	2
			d4279d9115abee44966bba07eba6bb8933eff7a8		
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National Stage of an International Application under 35 U.S.C. 371

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